

A. Structural bond: Forces which are more intense than those in the matrix
between the structure and the matrix

Interfacial: Forces which include chemical and physical
 forces and which are specific

cohesion: an atom force attraction to neighboring atoms

adhesion: an atom force attraction to foreign materials

B. Covalent compounds that have are arranged in a crystal lattice
structure atoms that are joined by

Properties of covalent compounds:

- High melting point low boiling point
- hard not very soluble in water
- soluble electrical insulator not conductor
usually insulate the current from the system

covalent bond: atoms are joined by sharing
electrons

Covalent compounds have atoms joined by
electrons together

Properties of covalent compounds:

- High melting point low boiling point
- Many insoluble compounds are soluble in water
- water not very good
- not very soluble in water

electrochemical: property that allows some strong attraction between
the two atoms

ionic compounds are highly electrochemically active systems which
include strong attraction between atoms which are joined
by strong and the high negative partial charge

Note Taking Guide Episode 501 Chemistry

Yicheng Fang



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The Athenaeum ,1885 The Chemical Trade Journal and Chemical Engineer ,1953 New York Magazine
,1993-01-18 New York magazine was born in 1968 after a run as an insert of the New York Herald Tribune and quickly made a place for itself as the trusted resource for readers across the country With award winning writing and photography covering everything from politics and food to theater and fashion the magazine s consistent mission has been to reflect back to its audience the energy and excitement of the city itself while celebrating New York as both a place and an idea

British Medical Journal ,1937 **Paperbound Books in Print** ,1971-07 **The Book Review Digest** ,2006
New York Times Saturday Review of Books and Art ,1970 **The Compact Edition of the Oxford English Dictionary** Sir James Augustus Henry Murray,1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933

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