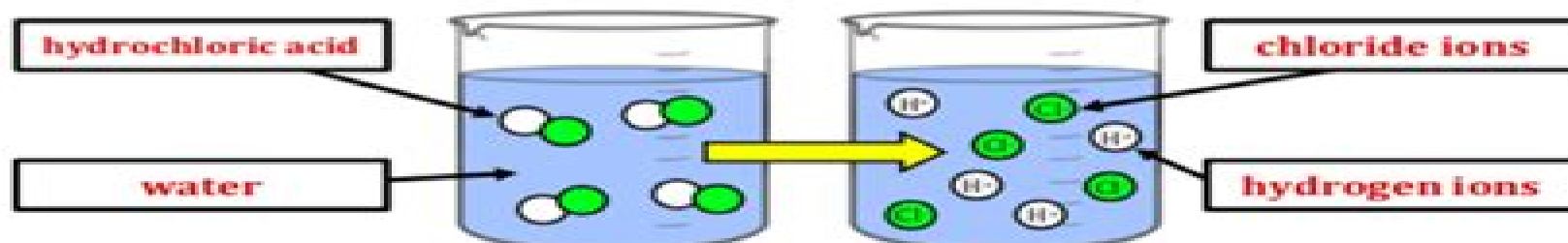


Introduction to Acids and Bases

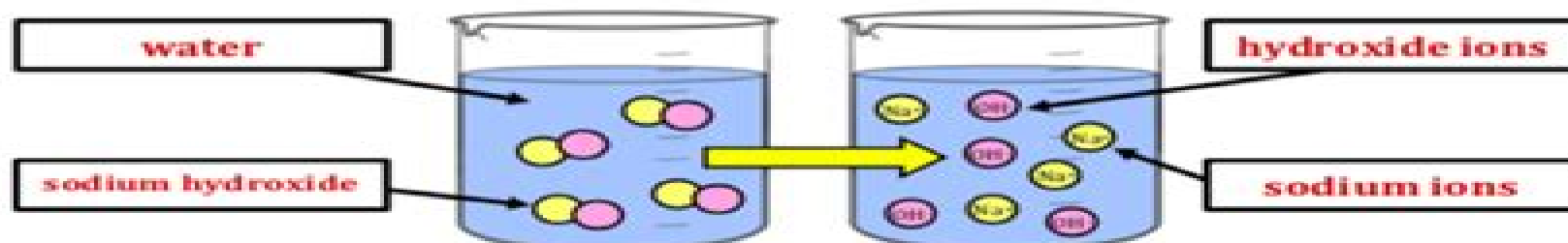
According to the **Arrhenius** definition:

- **Acids** are substances that release **hydrogen** ions in water (H^+)
- **Bases** are substances that release **hydroxide** ions in water (OH^-)

Instructions: The diagram below shows hydrochloric acid dissolving in water.
Label water, hydrochloric acid, hydrogen ions, and chloride ions.



Instructions: The diagram below shows sodium hydroxide dissolving in water.
Label water, sodium hydroxide, sodium ions, and hydroxide ions



Instructions: Identify whether the chemicals below would be *acids* or *bases*.

- | | |
|----------------------------|-------------|
| 1) HBr | <u>acid</u> |
| 2) LiOH | <u>base</u> |
| 3) H_2SO_4 | <u>acid</u> |

- | | |
|-----------------------------|-------------|
| 4) $\text{Ca}(\text{OH})_2$ | <u>base</u> |
| 5) HNO_3 | <u>acid</u> |
| 6) KOH | <u>base</u> |

Note Taking Worksheet Solutions Acids And Bases

T. R. Dickson



Note Taking Worksheet Solutions Acids And Bases:

Me n Mine-Science-Term-1 Saraswati Experts, A text book on science **Science on a Shoestring** Herb Strongin, Kara Strongin, Gloria Strongin, 1991 Hands on student investigations using low cost materials commonly found in local supermarkets **Basics of Analytical Chemistry and Chemical Equilibria** Brian M. Tissue, 2013-07-22 Enables students to progressively build and apply new skills and knowledge Designed to be completed in one semester this text enables students to fully grasp and apply the core concepts of analytical chemistry and aqueous chemical equilibria Moreover the text enables readers to master common instrumental methods to perform a broad range of quantitative analyses Author Brian Tissue has written and structured the text so that readers progressively build their knowledge beginning with the most fundamental concepts and then continually applying these concepts as they advance to more sophisticated theories and applications Basics of Analytical Chemistry and Chemical Equilibria is clearly written and easy to follow with plenty of examples to help readers better understand both concepts and applications In addition there are several pedagogical features that enhance the learning experience including Emphasis on correct IUPAC terminology You Try It spreadsheets throughout the text challenging readers to apply their newfound knowledge and skills Online tutorials to build readers skills and assist them in working with the text s spreadsheets Links to analytical methods and instrument suppliers Figures illustrating principles of analytical chemistry and chemical equilibria End of chapter exercises Basics of Analytical Chemistry and Chemical Equilibria is written for undergraduate students who have completed a basic course in general chemistry In addition to chemistry students this text provides an essential foundation in analytical chemistry needed by students and practitioners in biochemistry environmental science chemical engineering materials science nutrition agriculture and the life sciences Improving High School Students' Performance in Chemistry with a Hands-on Approach Mary L. Fredell, 1998

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vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

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