

Write Titrating Volume (Spreads: 1.00)

Name _____

Acetic Acid (pKa = 4.75) Titration with NaOH

a. $\text{pH} = 3.5$ volume NaOH (mL) = 10.0 (mL NaOH)

b. $\text{pH} = 4.75$ volume NaOH (mL) = 20.0 (mL NaOH)

c. $\text{pH} = 5.5$ (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH) (mL NaOH)

Acetic Acid (pKa = 4.75) Titration with NaOH

$\text{pH} = 3.5$ $\text{pH} = 4.75$ $\text{pH} = 5.5$
 $\text{pH} = 3.5$ $\text{pH} = 4.75$ $\text{pH} = 5.5$

Species	Add (mL) Base (NaOH or NaOH)	Acids with NaOH (NaOH or NaOH)	Conductivity (C or NaOH)
HCl	HCl	HCl	C
NaOH	NaOH	NaOH	C
NaCl	NaCl	NaCl	C
NaOH	NaOH	NaOH	C

Note Taking Guide 1101 Answers

**Micheál Ó Floinn, Lindsay Farmer, Julia
Hörnle, David Ormerod KC**

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