Question	Expected Answers	Marks	Additional Guidance
3 a	measured pH > 1 OR [H"] < 0.1 (mol dm ⁻³) <	4	ALLOW C ₂ H ₅ throughout question ALLOW [H*] < [CH ₃ CH ₂ COOH] OR [H*] < [HA] ALLOW measured pH is higher than expected ALLOW measured pH is not as acidic as expected ALLOW a quoted pH value or range > 1 and < 7 OR between 1 and 7
	[H'] = 10 ^{-6H} / K _a = [H'] CH ₃ CH ₃ COOT] OR BH'] ² / [CH ₃ CH ₃ COOH] Calculate K _a from [H'] ² / 0.100		ALLOW [H*] = antilog –pH OR [H*] = inverse log –pH ALLOW [H*][A*] OR [H*]* [HA] IF K_a is NOT given and $K_a = \frac{[H^a]^2}{0.100}$ is shown, award mark for K_a also (i.e. $K_a = \frac{[H^a]^2}{0.100}$ is automatically awarded the last 2 marks).
b	Marks are for correctly calculated values. Working shows how values have been derived. [H'] = $10^{-13.61} = 3.47 \times 10^{-64}$ (mol dm ⁻²) \checkmark [OHf] = $\frac{1.0 \times 10^{-64}}{3.47 \times 10^{-64}} = 0.29$ (mol dm ⁻²) \checkmark	2	ALLOW 3.467368505 × 10 ⁻¹⁴ and correct rounding to 3.5 × 10 ⁻¹⁴ ALLOW 0.28840315 and correct rounding to 0.29, i.e. ALLOW 0.288 ALLOW alternative approach using pOH: pOH = 14 - 13.46 = 0.54 ✓ [OH] = 10 ^{-0.54} = 0.29 (mol dm ⁻⁰) ✓ Correct answer gets BOTH marks

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Judd E. Hollander

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