

Chemistry Worksheet Class 11 on Chapter 1 Some Basic Concepts of Chemistry - Set 2

Q1. What is the mass of one atom of C – 12 in grams?

- a.) 1.992×10^{-23} g
- b.) 1.989×10^{-23} g
- c.) 1.892×10^{-23} g
- d.) 1.965×10^{-23} g

Q2. Which of the following laws states that the total mass of the reactants is equal to the total mass of the products in a chemical reaction?

- a.) Law of multiple proportions
- b.) Law of conservation of mass
- c.) Avogadro's law
- d.) Law of constant proportions

Q3. The law of multiple proportions was stated by:

- a.) Dalton
- b.) Proust
- c.) Avogadro
- d.) None of the above

Q4. The units $\text{kg m}^{-1} \text{s}^{-2}$ are of ____.

- a.) Force
- b.) Pressure
- c.) Acceleration
- d.) Velocity

Q5. The empirical formula of glucose is ____.

- a.) CH_2O
- b.) CHO_2
- c.) $\text{C}_2\text{H}_2\text{O}$
- d.) CH_2O_2

Q6. Express the following in the scientific notation:

- i.) 0.0048

Self Check Reading And Study Worksheet Chemistry

Chapter 1

DP Hallahan



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Self Check Reading And Study Workksheet Chemistry Chapter 1 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Self Check Reading And Study Workksheet Chemistry Chapter 1**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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