TYPES OF CHEMICAL REACTIONS

Identify each type of reaction:

- 1. Na₃ (PO₄) + 3 KOH --> 3 NaOH + K₃ (PO₄) double displacement
- MgCl₂ + Li₂(CO₃) --> Mg(CO₃) + 2 LiCl double displacement
- 3. C₆H₁₂ + 9 O₂ --> 6 CO₂ + 6 H₂O combustion
- 4. Pb + Fe(SO₄) --> Pb(SO₄) + Fe single displacement
- 5. 2 AgNO₃ + Cu --> Cu(NO₃)₂ + 2 Ag single displacement
- 6. C₃H₆O + 4 O₂ --> 3 CO₂ + 3 H₂O combustion
- 2 (C₅H₅) + Fe --> Fe(C₅H₅)₂
 synthesis
- 8. CaCO₃ --> CaO + CO₂ decomposition
- 9. P₄ + 3 O₂ --> 2 P₂O₃ synthesis
- 10. 2 RbNO₃ + BeF₂ --> Be(NO₃)₂ + 2 RbF double displacement

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