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**Exceed to Succeed**

## **Worksheet: Periodic Trends**

### **1. ATOMIC RADIUS**

For each of the following sets of atoms, rank the atoms from smallest to largest atomic radius.

- a. Li, C, F
- b. Li, Na, K
- c. Ge, P, O
- d. C, N, Al
- e. Al, Cl, Ga

### **2. IONIC RADIUS**

For each of the following sets of ions, rank them from smallest to largest ionic radius.

- a.  $\text{Mg}^{2+}$ ,  $\text{Si}^{4+}$ ,  $\text{S}^{2-}$
- b.  $\text{Mg}^{2+}$ ,  $\text{Ca}^{2+}$ ,  $\text{Ba}^{2+}$
- c.  $\text{F}^-$ ,  $\text{Cl}^-$ ,  $\text{Br}^-$
- d.  $\text{Ba}^{2+}$ ,  $\text{Cu}^{2+}$ ,  $\text{Zn}^{2+}$
- e.  $\text{Si}^{4+}$ ,  $\text{P}^{3-}$ ,  $\text{O}^{2-}$

### **3. IONIZATION ENERGY**

For each of the following sets of atoms, rank them from lowest to highest ionization energy.

- a. Mg, Si, S
- b. Mg, Ca, Ba
- c. F, Cl, Br
- d. Ba, Cu, Ne
- e. Si, P, He

### **4. ELECTRONEGATIVITY**

For each of the following sets of atoms, rank them from lowest to highest electronegativity.

- a. Li, C, N
- b. C, O, Ne
- c. Si, P, O
- d. K, Mg, P
- e. S, F, He

# Periodic Trends Extension Questions Worksheet

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Frew**



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*Merrill Chemistry* Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998      **Occupational Safety and Hygiene**  
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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Periodic Trends Extension Questions Worksheet**," a mesmerizing literary creation penned with a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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