

# Periodic Table of the Elements

1 IA <b>H</b> Hydrogen (1.008)																	18 VIIIA <b>He</b> Helium (4.003)				
3 IIA <b>Li</b> Lithium (6.941)	4 IIA <b>Be</b> Beryllium (9.012)															5 IIIA <b>B</b> Boron (10.81)	6 IIIA <b>C</b> Carbon (12.011)	7 IVA <b>N</b> Nitrogen (14.007)	8 IVA <b>O</b> Oxygen (15.999)	9 VA <b>F</b> Fluorine (18.998)	10 VA <b>Ne</b> Neon (20.180)
11 IA <b>Na</b> Sodium (22.990)	12 IIA <b>Mg</b> Magnesium (24.305)															13 IIIA <b>Al</b> Aluminum (26.982)	14 IIIA <b>Si</b> Silicon (28.086)	15 IVA <b>P</b> Phosphorus (30.974)	16 IVA <b>S</b> Sulfur (32.06)	17 VA <b>Cl</b> Chlorine (35.45)	18 VA <b>Ar</b> Argon (39.948)
19 IA <b>K</b> Potassium (39.098)	20 IIA <b>Ca</b> Calcium (40.078)	21 IIB <b>Sc</b> Scandium (44.956)	22 IIB <b>Ti</b> Titanium (47.88)	23 IIB <b>V</b> Vanadium (50.942)	24 IIB <b>Cr</b> Chromium (51.996)	25 IIB <b>Mn</b> Manganese (54.938)	26 IIB <b>Fe</b> Iron (55.845)	27 IIB <b>Co</b> Cobalt (58.933)	28 IIB <b>Ni</b> Nickel (58.693)	29 IIB <b>Cu</b> Copper (63.546)	30 IIB <b>Zn</b> Zinc (65.38)	31 IIIB <b>Ga</b> Gallium (69.723)	32 IIIB <b>Ge</b> Germanium (72.64)	33 IIIB <b>As</b> Arsenic (74.922)	34 IIIB <b>Se</b> Selenium (78.96)	35 IIIB <b>Br</b> Bromine (79.904)	36 IIIB <b>Kr</b> Krypton (83.798)				
37 IA <b>Rb</b> Rubidium (85.468)	38 IIA <b>Sr</b> Strontium (87.62)	39 IIB <b>Y</b> Yttrium (88.906)	40 IIB <b>Zr</b> Zirconium (91.224)	41 IIB <b>Nb</b> Niobium (92.906)	42 IIB <b>Mo</b> Molybdenum (95.94)	43 IIB <b>Tc</b> Technetium (98.906)	44 IIB <b>Ru</b> Ruthenium (101.07)	45 IIB <b>Rh</b> Rhodium (102.91)	46 IIB <b>Pd</b> Palladium (106.36)	47 IIB <b>Ag</b> Silver (107.87)	48 IIB <b>Cd</b> Cadmium (112.41)	49 IIIB <b>In</b> Indium (114.82)	50 IIIB <b>Sn</b> Tin (118.71)	51 IIIB <b>Sb</b> Antimony (121.76)	52 IIIB <b>Te</b> Tellurium (127.6)	53 IIIB <b>I</b> Iodine (126.91)	54 IIIB <b>Xe</b> Xenon (131.29)				
55 IA <b>Cs</b> Cesium (132.91)	56 IIA <b>Ba</b> Barium (137.33)	57-71 Lanthanides	72 IIIB <b>Hf</b> Hafnium (178.49)	73 IIIB <b>Ta</b> Tantalum (180.95)	74 IIIB <b>W</b> Tungsten (183.84)	75 IIIB <b>Re</b> Rhenium (186.21)	76 IIIB <b>Os</b> Osmium (190.23)	77 IIIB <b>Ir</b> Iridium (192.22)	78 IIIB <b>Pt</b> Platinum (195.08)	79 IIIB <b>Au</b> Gold (196.97)	80 IIIB <b>Hg</b> Mercury (200.59)	81 IIIB <b>Tl</b> Thallium (204.38)	82 IIIB <b>Pb</b> Lead (207.2)	83 IIIB <b>Bi</b> Bismuth (208.98)	84 IIIB <b>Po</b> Polonium (209)	85 IIIB <b>At</b> Astatine (210)	86 IIIB <b>Rn</b> Radon (222)				
87 IA <b>Fr</b> Francium (223)	88 IIA <b>Ra</b> Radium (226)	89-103 Actinides	104 IIIB <b>Rf</b> Rutherfordium (261)	105 IIIB <b>Db</b> Dubnium (262)	106 IIIB <b>Sg</b> Seaborgium (266)	107 IIIB <b>Bh</b> Bohrium (264)	108 IIIB <b>Hs</b> Hassium (277)	109 IIIB <b>Mt</b> Meitnerium (268)	110 IIIB <b>Ds</b> Darmstadtium (271)	111 IIIB <b>Rg</b> Roentgenium (272)	112 IIIB <b>Cn</b> Copernicium (285)	113 IIIB <b>Nh</b> Nihonium (284)	114 IIIB <b>Fl</b> Flerovium (289)	115 IIIB <b>Mc</b> Moscovium (288)	116 IIIB <b>Lv</b> Livermorium (293)	117 IIIB <b>Ts</b> Tennessine (289)	118 IIIB <b>Og</b> Oganesson (294)				

Atomic Number →

→ Symbol

→ Name

→ Atomic Weight

→ Electrons per shell

1

H

Hydrogen

1.008

1

57 La Lanthanum (138.91)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.91)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.96)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.93)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.93)	68 Er Erbium (167.26)	69 Tm Thulium (168.93)	70 Yb Ytterbium (173.05)	71 Lu Lutetium (174.967)
89 Ac Actinium (227)	90 Th Thorium (232.04)	91 Pa Protactinium (231.04)	92 U Uranium (238.03)	93 Np Neptunium (237.05)	94 Pu Plutonium (244.06)	95 Am Americium (243.06)	96 Cm Curium (247.07)	97 Bk Berkelium (247.07)	98 Cf Californium (251.08)	99 Es Einsteinium (252.08)	100 Fm Fermium (257.10)	101 Md Mendelevium (258.11)	102 No Nobelium (259.10)	103 Lr Lawrencium (262.10)

# Periodic Table Section 3

**Disha Experts**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Disha Experts' text.

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**Descriptive Inorganic Chemistry, Third Edition** Geoff Rayner-Canham,Tina Overton,2003 For lower division courses with an equal balance of description and theory

Student Solutions Manual to Accompany General Chemistry Carole H. McQuarrie,2011-06-15 For years Donald McQuarrie s chemistry textbooks have been famous among students and professors alike for their wonderful problems The Solutions Manual to Accompany General Chemistry Fourth Edition lists even numbered chapter ending problems from the textbook and goes on to provide detailed solutions For students studying independently or in groups this solutions manual will be tremendously useful to help students perfect their problem solving skills and to master the covered concepts For years Donald McQuarrie s chemistry textbooks have been famous among students and professors alike for their wonderful problems The Solutions Manual to Accompany General Chemistry Fourth Edition lists even numbered chapter ending problems from the textbook and goes on to provide detailed solutions For students studying independently or in groups this solutions manual will be tremendously useful to help students perfect their problem solving skills and to master the covered concepts

**Understanding General Chemistry** Atef Korchef,2022-03-07 Understanding General Chemistry details the fundamentals of general chemistry through a wide range of topics relating the structure of atoms and molecules to the properties of matter Written in an easy to understand format with helpful pedagogy to fuel learning the book features main objectives at the beginning of each chapter get smart sections and check your reading section at the end of each chapter The text is filled with examples and practices that illustrate the concepts at hand In addition a summary and extensive MCQs exercises and problems with the corresponding answers and explanations are readily available Additional features include Alerts students to common mistakes and explains in simple ways and clear applications how to avoid these mistakes Offers answers and comments alongside sample problems enabling students to self evaluate their skill level Includes powerful methods easy steps simple and accurate interpretations and engaging applications to help students understand complex principles Provides a bridge to more complex topics such as solid state chemistry organometallic chemistry chemistry of main group elements inorganic chemistry and physical chemistry This introductory textbook is ideal for chemistry courses for non science majors as well as health sciences and preparatory engineering students

**All In One Chemistry ICSE Class 9 2021-22** Shikha Goel,Saleha Parvez,2021-07-17 1 All in One ICSE self study guide deals with Class 9 Chemistry 2 It Covers Complete Theory Practice Assessment 3 The Guide has been divided in 8 Chapters 4 Complete Study Focused Theories

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*Chemistry Essentials For Dummies* John T. Moore, 2019-04-15  
 Chemistry Essentials For Dummies 9781119591146 was previously published as Chemistry Essentials For Dummies 9780470618363 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product Whether studying chemistry as part of a degree requirement or as part of a core curriculum students will find Chemistry Essentials For Dummies to be an invaluable quick reference guide to the fundamentals of this often challenging course Chemistry Essentials For Dummies contains content focused on key topics only with discrete explanations of critical concepts taught in a typical two semester high school chemistry class or a college level Chemistry I course from bonds and reactions to acids bases and the mole This guide is also a perfect reference for parents who need to review critical chemistry concepts as they help high school students with homework assignments as well as for adult learners headed back into the classroom who just need to a refresher of the core concepts The Essentials For Dummies Series Dummies is proud to present our new series The Essentials For Dummies Now students who are prepping for exams preparing to study new material or who just need a refresher can have a concise easy to understand review guide that covers an entire course by concentrating solely on the most important concepts From algebra and chemistry to grammar and Spanish our expert authors focus on the skills students most need to succeed in a subject

*Structuralist Knowledge Representation* , 2022-05-16

**Organic Chemistry** Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps

students fully grasp the essential concepts at the root of organic chemistry Providing a practical learning experience with numerous opportunities for self testing the book contains Checklists of what students need to know before they begin to study a topic Checklists of concepts to be fully understood before moving to the next subject area Homework problems directly tied to each concept at the end of each chapter Embedded problems with answers throughout the material Experimental details and mechanisms for key reactions The reactions and mechanisms contained in the book describe the most fundamental concepts that are used in industry biological chemistry and biochemistry molecular biology and pharmacy The concepts presented constitute the fundamental basis of life processes making them critical to the study of medicine Reflecting this emphasis most chapters end with a brief section that describes biological applications for each concept This text provides students with the skills to proceed to the next level of study offering a fundamental understanding of acids and bases applied to organic transformations and organic molecules

**Physical Science in the Modern World** Jerry Marion, 2012-12-02 Physical Science in the Modern World surveys the whole range of the non biological sciences This book explores the significant ideas and concepts in chemistry physics astronomy geology and meteorology with emphasis on how these sciences bear strongly upon one another and how the basic principles are applied to each Organized into three part encompassing 29 chapters this book starts with an overview of the fundamental building blocks of matter and explains how they are assembled to form molecules rocks minerals and the Earth This text then examines the basic concepts of physical science by exploring the fundamental principles that govern all physical processes and we see how they relate to various everyday occurrences Other chapters consider how modern chemistry affects the world we live in and explain how the development of semiconductor materials has led in the development of miniature electronics This book is a valuable resource for physicists chemists astronomers geologists and meteorologists

**A New Approach to ICSE Chemistry for Class X (A.Y. 2023-24) Onward** V. K. Sally, 2023-05-20 A New Approach to ICSE Chemistry for Class X has been revised in accordance with the latest syllabus prescribed by the Council for the Indian School Certificate Examination New Delhi for Class X The main strength of the book lies in the presentation of scientific content which has been so arranged that the topics are linked with each other and do not cause any unnecessary burden on the mind of the student Salient features of this book are as follows Clear simple and easy language A large number of chemical reactions along with experimental observations Full sized diagrams as expected from a student in the examination Topic wise Video Lectures as a support for effective learning Concepts supplemented by suitable day to day examples Periodic Table showing mass number atomic number of various elements along with the list of actual names of the elements Highlighted important terms and definitions Well designed exercises to assess conceptual reasoning skill and application based learning as per the latest recommendations of ICSE board Solved numerical problems at the end of each chapter to help the students solve numericals on their own A chapter on Practical Chemistry to help students in their laboratory work The latest Solved ICSE Specimen Paper has been given Scan

QR codes given at the end of each chapter to get the solution of chapter wise ICSE Board Examination Questions We hope this book will prove useful to fellow teachers and students Suggestions for further improvement of this book shall be gratefully acknowledged Author *Super 10 CBSE Class 12 Chemistry 2021-22 Term I Sample Papers with OMR Sheets* Disha Experts,2021-09-01 **Foundation Course in Chemistry for JEE/ NEET/ Olympiad Class 10 with Case Study Approach - 5th Edition** Disha Experts,2020-04-06 Foundation Course in Chemistry for JEE NEET Olympiad Class 10 with Case Study Approach is the thoroughly revised and updated 5th edition 2 colour of the comprehensive book for Class 10 students who aspire to become Doctors Engineers The book is focused at 3 Goals Bring Concept Clarity Sharpen Problem Solving Build a Strong Foundation The book discusses theoretical concepts in detail accompanied by Illustrations Learn More Let s Do Activity Did You Know Time to Check your Knowledge Another unique feature of this book is the Case Study Approach where most critical Problem Solving Concepts are discussed in various Permutations and Combinations so as improve Problem Solving Skills among the students The theory is followed by the Exercise part which covers in total 1800 questions divided into 4 levels of fully solved exercises which are graded as per their level of difficulty Exercise 1 Master Boards MCQs FIB True False Assertion Reason Passage Matching Very Short Short Long Answer Type Questions including Past Years Board Qns This Exercise also includes Reasoning Based HOTS and Case Based MCQs Exercise 2 Master the NCERT All Textbook Exemplar Questions Exercise 3 Foundation Builder Question Bank on NCERT chapter including MCQs 1 Correct MCQs 1 Correct Passage Assertion Reason Multiple Matching and Numeric Integer Type Questions with past years NTSE JSTSE KVPY NEET JEE Main considering Syllabus and Level of difficulty Exercise 4 Foundation Builder Question Bank on Connecting Topics Chapters including MCQs 1 Correct MCQs 1 Correct Passage Assertion Reason Multiple Matching and Numeric Integer Type Questions with past years NTSE JSTSE KVPY NEET JEE Main considering Syllabus and Level of difficulty The book adheres to the latest syllabus set by the NCERT going beyond by incorporating those topics which will assist the students to scale up in the next classes to achieve their academic dreams of Medicine or Engineering **Oswaal CBSE Question Bank Class 11 Chemistry, Chapterwise and Topicwise Solved Papers For 2025 Exams** Oswaal Editorial Board,2024-02-03 Description of the product 100% Updated Syllabus Question Typologies We have got you covered with the latest and 100% updated curriculum along with the latest typologies of Questions Timed Revision with Topic wise Revision Notes Smart Mind Maps Study smart not hard Extensive Practice with 1000 Questions SAS Questions Sri Aurobindo Society To give you 1000 chances to become a champ Concept Clarity with 500 Concepts Concept Videos For you to learn the cool way with videos and mind blowing concepts NEP 2020 Compliance with Competency Based Questions Artificial Intelligence For you to be on the cutting edge of the coolest educational trends *Basic Concepts of Chemistry* Leo J. Malone,Theodore O. Dolter,2011-12-27 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and

advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text *Educart ICSE Class 10 CHEMISTRY Question Bank + Solved Papers (10 Years) for 2023-2024* Educart,2023-05-27

**Basic Concepts Of Inorganic Chemistry** Singh,2009-09 **Essence of Physics, Chemistry, and Mathematics** Dr. Akhilesh Sharma,S.B. Singh,Yogendra Mohan,2024-04-17 This book has three sections namely Physics Chemistry and Mathematics having 15 6 and 13 chapters respectively with illustrations The book contains the previous 5 years question papers year wise with solutions NEE I The book is useful for NEE I and other competitive exams covering Physics Chemistry and Mathematics The book is relevant easy to understand

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