

ALKALI METALS

ALKALINE EARTH METALS

TRANSITION METALS

LANTHANIDES AND ACTINIDES

The image shows a standard periodic table of elements, color-coded to categorize them. Three callout boxes highlight specific groups:

- ALKALINE EARTH METALS:** A blue box points to the second column (Group 2), which includes elements like Beryllium (Be), Magnesium (Mg), Calcium (Ca), Strontium (Sr), Barium (Ba), and Radium (Ra).
- TRANSITION METALS:** A yellow box points to the d-block, which includes elements from Groups 3 to 10, such as Scandium (Sc), Titanium (Ti), Iron (Fe), Nickel (Ni), and Copper (Cu).
- LANTHANIDES AND ACTINIDES:** A light blue box points to the two rows of elements at the bottom of the table, which are the f-block elements. The top row contains Lanthanides (La to Lu) and the bottom row contains Actinides (Ac to Lr).

The periodic table is organized by atomic number (1 to 118) and grouped into 18 vertical columns (Groups I to VIII) and 7 horizontal rows (Periods 1 to 7). The colors used are: pink for alkali metals (Group 1), blue for alkaline earth metals (Group 2), yellow for transition metals (Groups 3-10), orange for post-transition metals (Groups 11-12), green for nonmetals (Groups 13-16), red for halogens (Group 17), and light blue for noble gases (Group 18).

Periodic Table Physical Properties Study

Page Keeley, Joyce Tugel



Periodic Table Physical Properties Study:

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