

2.6 - Periodic Trends worksheet

Name _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table

_____ 1. What elements are in period 5?

What element(s) is a

_____ 2. halogens

_____ 3. lanthanides

_____ 4. alkaline earth metals

_____ 5. noble gases

_____ 6. transition metals

_____ 7. alkali metals

_____ 8. actinides

What elements are in the sublevel

_____ 9. s

_____ 10. p

_____ 11. d

_____ 12. f

What elements are

_____ 13. metals,

_____ 14. nonmetals

_____ 15. metalloids,

_____ 16. Which element is the most reactive nonmetal?

_____ 17. Which element is the most reactive metal?

_____ 18. Which element below is the least reactive?

_____ 19. Which element has the largest radius in period 4?

_____ 20. Which element has the largest radius in the Alkali metals family?

_____ 21. Which element has the highest ionization energy in period four?

_____ 22. Which elements has the second largest atomic number?

_____ 23. Which element has the largest atomic mass?

_____ 24. Which element is located at period 4, group 15?

_____ 25. Which element is located at period 5, group 2?

_____ 27. Which elements will form +1 ions?

_____ 28. Which group of metals will have the same properties

_____ 29. Which group of nonmetals will have the same properties

_____ 30. On the back, define all chapter 5, section 3 vocabulary terms.

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