

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.008	2 He Helium 4.003																
3 Li Lithium 6.941	4 Be Beryllium 9.012																
5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180												
11 Na Sodium 22.990	12 Mg Magnesium 24.305																
13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948												
19 K Potassium 39.098	20 Ca Calcium 40.078																
21 Sc Scandium 44.956	22 Ti Titanium 47.867	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.38								
31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.798												
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29
55 Cs Cesium 132.905	56 Ba Barium 137.327																
57 Pr Praseodymium 140.908	58 Nd Neodymium 144.24																
59 Pm Promethium 144.913	60 Ce Cerium 140.12	61 La Lanthanum 138.905	62 Pr Praseodymium 140.908	63 Nd Neodymium 144.24	64 Pm Promethium 144.913	65 Sm Samarium 150.36	66 Eu Europium 151.964	67 Gd Gadolinium 157.25	68 Tb Terbium 158.925	69 Dy Dysprosium 162.50	70 Ho Holmium 164.930	71 Er Erbium 167.259	72 Tm Thulium 168.934	73 Yb Ytterbium 173.054	74 Lu Lutetium 174.967		
75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222	87 Fr Francium 223	88 Ra Radium 226	89 Ac Actinium 227	90 Th Thorium 232.038		
93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.083	99 Es Einsteinium 252.083	100 Fm Fermium 257.103	101 Md Mendelevium 258.103	102 No Nobelium 259	103 Lr Lawrencium 262							

Atomic #

Symbol

Full name

Weight

C Solid

Hg Liquid

H Gas

Rf Unknown

Metals

Other nonmetals

Nonmetals

Hydrogens

Noble gases

Alkali metals

Alkaline earth metals

Lanthanoids

Transition metals

Post-transition metals

1

2

3

4

5

6

7

8

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14

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18

273

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (same rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.



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Michal Rosen-Zvi



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Table of Contents Periodic Table Basic Speedy Study Guide Speedy Publishing

1. Understanding the eBook Periodic Table Basic Speedy Study Guide Speedy Publishing
 - The Rise of Digital Reading Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Basic Speedy Study Guide Speedy Publishing
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Personalized Recommendations
 - Periodic Table Basic Speedy Study Guide Speedy Publishing User Reviews and Ratings
 - Periodic Table Basic Speedy Study Guide Speedy Publishing and Bestseller Lists
5. Accessing Periodic Table Basic Speedy Study Guide Speedy Publishing Free and Paid eBooks
 - Periodic Table Basic Speedy Study Guide Speedy Publishing Public Domain eBooks
 - Periodic Table Basic Speedy Study Guide Speedy Publishing eBook Subscription Services
 - Periodic Table Basic Speedy Study Guide Speedy Publishing Budget-Friendly Options

6. Navigating Periodic Table Basic Speedy Study Guide Speedy Publishing eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Basic Speedy Study Guide Speedy Publishing Compatibility with Devices
 - Periodic Table Basic Speedy Study Guide Speedy Publishing Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Highlighting and Note-Taking Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Interactive Elements Periodic Table Basic Speedy Study Guide Speedy Publishing
8. Staying Engaged with Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Basic Speedy Study Guide Speedy Publishing
9. Balancing eBooks and Physical Books Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Basic Speedy Study Guide Speedy Publishing
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Setting Reading Goals Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Fact-Checking eBook Content of Periodic Table Basic Speedy Study Guide Speedy Publishing
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

-
- Interactive and Gamified eBooks

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~~recipe for milk chocolate mousse~~

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