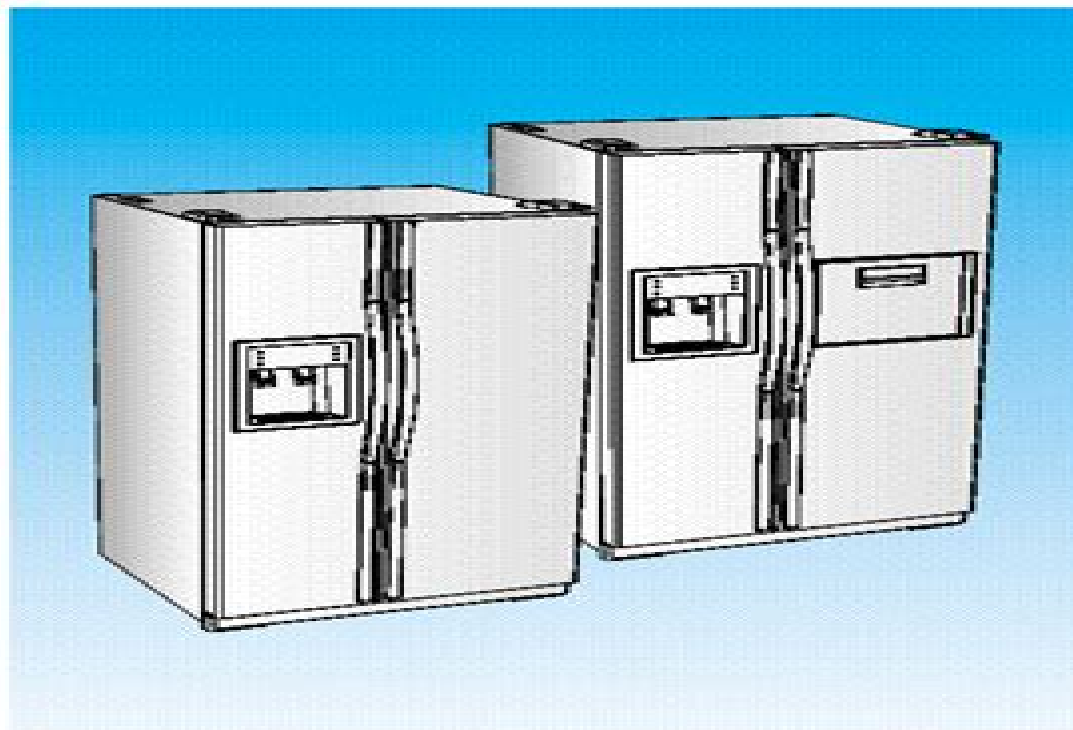


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# ***SERVICE GUIDE***

**SIDE-BY-SIDE REFRIGERATOR**



**Model:  
RS2555SL**

**SAM0061**

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**Whirlpool Corporation**



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web atoms of the same element i e same atomic number  $Z$  that have different numbers of neutrons are called isotopes for example 99 of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web may 16 2018 explanation an atom is characterized as a particular element by the number of positively charged massive nuclear particles i.e.  $Z$   $Z=1$  we got hydrogen  $Z=2$  we got helium  $Z=3$  lithium  $Z=47$  we got silver the nucleus can also contain neutrons massive nuclear particles of zero charge

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web oct 31 2023 an isotope is one of two or more species of atoms of a chemical element with the same atomic number and

position in the periodic table and nearly identical chemical behavior but with different atomic masses and physical properties every chemical element has one or more isotopes

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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web nov 27 2016 explanation basically isotopes are nuclei that have the same number of protons and are thus the same element but have unequal numbers of neutrons answer link the nucleus differs as a result of the number of neutrons in the atom the actual structure doesn t differ all that much but its mass does differ thus its density is a little

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