#### Topics:

- 1. General Chemistry
- 2. Molecular Representations
- 3. Nomenclature
- 4. Alkanes/Cycloalkanes
- 5. Stereoisomers
- 6. Chemical reactivity/mechanism
- 7. SN/E Reactions

#### General Chemistry Questions

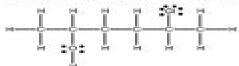
- 1. Identify all of the intermolecular forces present in each molecule:
  - a. CH<sub>2</sub>OH
  - b. CH<sub>2</sub>COCH<sub>2</sub>CH<sub>3</sub>
  - c. CH<sub>2</sub>C(CH<sub>3</sub>)<sub>2</sub>
- 2. Fill out the blanks in the chart (VESPR Theory)

| Hybridization   | Geometry    | # of Bonds   | # of Lone Pairs | Bond Angle        |
|-----------------|-------------|--|-----------------|-------------------|
|                 | Tetrahedral | The state of the s | 0               | The second second |
| Sp <sup>3</sup> |             | 3  | 1               |                   |
|                 | Bent        |  |                 | 109.5             |
| Sp <sup>2</sup> | 3,550       | 3  | 0               |                   |
|                 |             | 2  | 1               | 120               |
|                 |             |  | 0               | 180               |

- 3. Determine if the following molecules have a dipole moment
  - a. CH<sub>2</sub>CH<sub>3</sub>
  - b. CH3CH2CH(NH<sub>2</sub>)CH<sub>3</sub>
  - c. CH<sub>2</sub>Cl<sub>2</sub>
  - d. H<sub>2</sub>O\*
  - e. HOOst

#### Molecular Representations Questions:

1. Write the condensed name for the following structures:



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