

**Max. Marks:70**

**Time: 3 hours**

**General Instructions:**

**Read the following instructions carefully.**

- (a) There are 33 questions in this question paper with internal choice.
  - (b) SECTION A consists of 16 multiple -choice questions carrying 1 mark each.
  - (c) SECTION B consists of 5 short answer questions carrying 2 marks each.
  - (d) SECTION C consists of 7 short answer questions carrying 3 marks each.
  - (e) SECTION D consists of 2 case - based questions carrying 4 marks each.
  - (f) SECTION E consists of 3 long answer questions carrying 5 marks each.
  - (g) All questions are compulsory.
  - (h) Use of log tables and calculators is not allowed.
- 

### SECTION A

*The following questions are multiple -choice questions with one correct answer. Each question carries 1 mark. There is no internal choice in this section.*

1. Which of the following solutions will have the highest conductivity at 298 K?
 

(a) 0.01 M HCl solution	(b) 0.1 M HCl solution
(c) 0.01 M CH <sub>3</sub> COOH solution	(d) 0.1 M CH <sub>3</sub> COOH solution



Identify A and B:

- |  |   |
|--|---|
| (a) A = 1-phenylethanal , B = acetophenone | (b) A = Benzophenone B = formaldehyde   |
| (c) A = Benzaldehyde , B = Acetophenone    | (d) A = Benzophenone , B = Acetophenone |
3. The vitamins which can be stored in our body are:
 

(a) Vitamin A, B, D and E	(d) Vitamin A, C, D and K
(c) Vitamin A, B, C and D	(d) Vitamin A, D, E and K
  4. What is IUPAC name of the ketone A, which undergoes iodo form reaction to give CH<sub>3</sub>CH= C(CH<sub>3</sub>)COONa and yellow precipitate of CHI<sub>3</sub>?
 

(a) 3-Methylpent-3-en-2-one	(b) 3-Methylbut-2-en- one
(c) 2, 3-Dimethylethanone	(d) 3-Methylpent-4-one
  5. Which of the following is not correct?
 

(a) In haloarenes, the electron pairs on halogen atom are in conjugation with $\pi$ -electrons of the ring.
(b) The carbon-magnesium bond is covalent and non-polar in nature .
(c) During S <sub>N</sub> <sup>1</sup> reaction, the carbocation formed in the slow step being sp <sup>2</sup> hybridised is planar.
(d) Out of CH <sub>2</sub> = CH-Cl and C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Cl, C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> Cl is more reactive towards S <sub>N</sub> <sup>1</sup> reaction

# Sample Question Paper Applied Chemistry

**JE Gale**



## **Sample Question Paper Applied Chemistry:**

This book delves into Sample Question Paper Applied Chemistry. Sample Question Paper Applied Chemistry is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Sample Question Paper Applied Chemistry, encompassing both the fundamentals and more intricate discussions.

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- Chapter 3: Sample Question Paper Applied Chemistry in Everyday Life
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## **Table of Contents Sample Question Paper Applied Chemistry**

1. Understanding the eBook Sample Question Paper Applied Chemistry
  - The Rise of Digital Reading Sample Question Paper Applied Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Sample Question Paper Applied Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Sample Question Paper Applied Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Sample Question Paper Applied Chemistry
  - Personalized Recommendations
  - Sample Question Paper Applied Chemistry User Reviews and Ratings
  - Sample Question Paper Applied Chemistry and Bestseller Lists
5. Accessing Sample Question Paper Applied Chemistry Free and Paid eBooks
  - Sample Question Paper Applied Chemistry Public Domain eBooks
  - Sample Question Paper Applied Chemistry eBook Subscription Services
  - Sample Question Paper Applied Chemistry Budget-Friendly Options
6. Navigating Sample Question Paper Applied Chemistry eBook Formats
  - ePub, PDF, MOBI, and More
  - Sample Question Paper Applied Chemistry Compatibility with Devices
  - Sample Question Paper Applied Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Sample Question Paper Applied Chemistry
  - Highlighting and Note-Taking Sample Question Paper Applied Chemistry
  - Interactive Elements Sample Question Paper Applied Chemistry

8. Staying Engaged with Sample Question Paper Applied Chemistry
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Sample Question Paper Applied Chemistry
9. Balancing eBooks and Physical Books Sample Question Paper Applied Chemistry
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Sample Question Paper Applied Chemistry
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Sample Question Paper Applied Chemistry
  - Setting Reading Goals Sample Question Paper Applied Chemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Sample Question Paper Applied Chemistry
  - Fact-Checking eBook Content of Sample Question Paper Applied Chemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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