

- Q6 When water was added to the anhydrate of  $\text{CuSO}_4(x)$ , did the reaction proceed in the direction of the reactants or products?

## B.2 Reaction of Copper(II) Ion and Hydroxide

|                                               | Test tube 1       | Test tube 2      |
|-----------------------------------------------|-------------------|------------------|
| Initial color of $\text{CuCl}_2$ solution (1) | light blue        | light blue       |
| Initial appearance $\text{Cu(OH)}_2(s)$ (2)   | clear             | clear            |
| Change in appearance (3)                      | cloudy            | <del>clear</del> |
| Change in appearance (4)                      | <del>cloudy</del> | clear            |

- Q7 In test tube 1, did the addition of  $\text{OH}^-$  cause the reaction to go in the direction of the reactants or products? Explain.
- Q8 In test tube 2, did the addition of  $\text{H}_3\text{O}^+$  cause the reaction to go in the direction of the reactants or products? Explain.

## C. Changing Equilibrium Conditions: Le Châtelier's Principle

### C.1 Effect of Changing Reactant Concentration

| Test tube                   | Color      | Is color deeper or lighter red compared to the reference? | $[\text{FeSCN}^{2+}]$ increases or decreases? |
|-----------------------------|------------|-----------------------------------------------------------|-----------------------------------------------|
| 1. Reference                | red-orange | <del>deeper/lighter</del>                                 | <del>increases/decreases</del>                |
| 2. $\text{Fe}^{3+}$ added   | dark red   | deeper                                                    | increased                                     |
| 3. $\text{SCN}^-$ added     | blood red  | deeper                                                    | increased                                     |
| 4. $\text{Fe}^{3+}$ removed | orange     | lighter                                                   | decreased                                     |

# Reaction Rates And Equilibrium Lab Answers

**Mina L. Carnicom**



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**Safety-Scale Lab Exp Biochem 2e** Spencer L. Seager, Michael R. Slabaugh, 1994-05

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data is supported by the Data Analysis section      **Geochemical Modeling for Mine Site Characterization and Remediation** D. Kirk Nordstrom, Andrew Nicholson, 2017-10-01 The single most important factor for the successful application of a geochemical model is the knowledge and experience of the individuals conducting the modeling Geochemical Modeling for Mine Site Characterization and Remediation is the fourth of six volumes in the Management Technologies for Metal Mining Influenced Water series about technologies for management of metal mine and metallurgical process drainage This handbook describes the important components of hydrogeochemical modeling for mine environments primarily those mines where sulfide minerals are present metal mines and coal mines It provides general guidelines on the strengths and limitations of geochemical modeling and an overview of its application to the hydrogeochemistry of both unmined mineralized sites and those contaminated from mineral extraction and mineral processing The handbook includes an overview of the models behind the codes explains vital geochemical computations describes several modeling processes provides a compilation of codes and gives examples of their application including both successes and failures Hydrologic modeling is also included because mining contaminants most often migrate by surface water and groundwater transport and contaminant concentrations are a function of water residence time as well as pathways This is an indispensable resource for mine planners and engineers environmental managers land managers consultants researchers government regulators nongovernmental organizations students stakeholders and anyone with an interest in mining influenced water The other handbooks in the series are Basics of Metal Mining Influenced Water Mitigation of Metal Mining Influenced Water Mine Pit Lakes Characteristics Predictive Modeling and Sustainability Techniques for Predicting Metal Mining Influenced Water and Sampling and Monitoring for the Mine Life Cycle      *Chemistry* McGraw-Hill Staff, 2001-07      *ERDA Energy Research Abstracts* ,1983      [Subject Index to Unclassified ASTIA Documents](#) Defense Documentation Center (U.S.), 1960      **Solar Energy Update** ,1982      **The Software Encyclopedia 2001** ,2001      **Government Reports Announcements & Index** ,1992-09      **Aero/space Engineering** ,1958

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