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Student Exploration: Reaction Energy

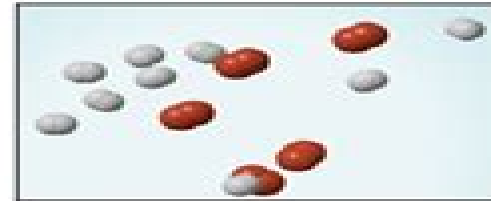
Vocabulary: calorimeter, chemical bond, endothermic, enthalpy, exothermic, Hess's law

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Two magnets are stuck together. What might you have to do to get them to separate?
Pull them apart
2. Suppose you held two magnets a short distance apart, then let go. What would happen?
They would connect
3. Think about the magnets in terms of energy. In which case do you increase the potential energy of the magnets? In which case do you increase the kinetic energy of the magnets?

Gizmo Warm-up

Just like magnets, atoms of different elements are attracted together to form **chemical bonds**. Breaking these bonds requires energy. When a new bond forms, energy is released and temperatures rise. In the *Reaction Energy Gizmo™*, you will explore how the energy of chemical bonding relates to temperature changes that occur during chemical reactions.



To begin, check that **Reaction 1** and **Forward** are selected. In this reaction, hydrogen (H_2) and oxygen (O_2) react to form water (H_2O). The reaction takes place inside a device called a **calorimeter**. Inside the calorimeter, a small chamber holds the reactants. The rest of the calorimeter is filled with water.

1. Click **Play** (▶). What happens? When I clicked play, all of the separate H_2 and O_2 molecules formed to create H_2O .
2. How does the temperature change? The temperature is increased while the molecules are reacting to form water, and the increase in temperature slows once the molecules are all bonded to create the H_2O molecules. The temperature raises from 21.0 Degrees Celsius to 26.0 Degrees Celsius while the molecules were bonding.

Reaction Energy Study Guide Answers

Rachel Sandford



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