# **Predicting Reaction Products**

Predict the products for the following reactions, balance the equation, then classify the type of reaction:

- Na + \_\_\_ FeBr<sub>3</sub> →
- 2) \_\_\_\_ NaOH + \_\_\_\_ H<sub>2</sub>SO<sub>4</sub> →
- 3) \_\_\_\_ C<sub>2</sub>H<sub>4</sub>O<sub>2</sub> + \_\_\_\_ O<sub>2</sub> →
- 4) \_\_\_\_ NH<sub>3</sub> + \_\_\_\_ H<sub>2</sub>O →
- 5) \_\_\_\_ PbSO<sub>4</sub> + \_\_\_\_ AgNO<sub>3</sub> →
- 6) \_\_\_\_\_PBr₃ →
- 7) HBr + Fe →
- 8) \_\_\_\_ KMnO<sub>4</sub> + \_\_\_ ZnCl<sub>2</sub> →
- 9) \_\_\_\_MnO<sub>2</sub> + \_\_\_\_ Sn(OH)<sub>4</sub> →
- 10) \_\_\_\_ O<sub>2</sub> + \_\_\_\_ C<sub>3</sub>H<sub>12</sub>O<sub>2</sub> →
- 11) \_\_\_\_ H<sub>2</sub>O<sub>2</sub> →
- 12) \_\_\_\_ PtCl<sub>4</sub> + \_\_\_ Cl<sub>2</sub> →

# **Reaction Products Worksheet Chemfiesta**

F Rizvi

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