

Fictitious symbols are used for the first 18 elements in the periodic table. Use the clues below to write the fictitious symbol in the appropriate spot on the periodic table provided. Symbols for real elements do not represent those elements. HINT: You do not have to complete each clue in order.

1	2	13	14	15	16	17	18
X		A	T	H P	G	Q	L
U	R	AI	DP	Z	E	K	W
J	M						V

- Clue #1 U and J are alkali metals. J has more energy levels than U.
- Clue #2 I has 4 valence electrons on the ^{2nd} ~~3rd~~ energy level.
- Clue #3 M is a metal in period 3 with 2 valence electrons
- Clue #4 X has one proton in its nucleus
- Clue #5 Q has 2 energy levels and is a halogen
- Clue #6 L is the only noble gas that doesn't have 8 valence electrons
- Clue #7 A is the only metalloid in the 13th group.
- Clue #8 I has 3 valence electrons on the third energy level
- Clue #9 V and W are un-reactive elements. V has more electrons than W.
- Clue #10 E and G have a common ion of -2 (E^{-2} , G^{-2}). E has 16 neutrons.
- Clue #11 R has an nuclear symbol of 3_1R
- Clue #12 K has one less electron than V
- Clue #13 D and Z are in the same period as J. However D is a metalloid and Z is a nonmetal.
- Clue #14 H has 7 neutrons

Periodic Trends Puzzle Answer Key

CO Houle



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Table of Contents Periodic Trends Puzzle Answer Key

1. Understanding the eBook Periodic Trends Puzzle Answer Key
 - The Rise of Digital Reading Periodic Trends Puzzle Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Puzzle Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Puzzle Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Puzzle Answer Key
 - Personalized Recommendations
 - Periodic Trends Puzzle Answer Key User Reviews and Ratings
 - Periodic Trends Puzzle Answer Key and Bestseller Lists

5. Accessing Periodic Trends Puzzle Answer Key Free and Paid eBooks
 - Periodic Trends Puzzle Answer Key Public Domain eBooks
 - Periodic Trends Puzzle Answer Key eBook Subscription Services
 - Periodic Trends Puzzle Answer Key Budget-Friendly Options
6. Navigating Periodic Trends Puzzle Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Puzzle Answer Key Compatibility with Devices
 - Periodic Trends Puzzle Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Puzzle Answer Key
 - Highlighting and Note-Taking Periodic Trends Puzzle Answer Key
 - Interactive Elements Periodic Trends Puzzle Answer Key
8. Staying Engaged with Periodic Trends Puzzle Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Puzzle Answer Key
9. Balancing eBooks and Physical Books Periodic Trends Puzzle Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Puzzle Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Puzzle Answer Key
 - Setting Reading Goals Periodic Trends Puzzle Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Puzzle Answer Key
 - Fact-Checking eBook Content of Periodic Trends Puzzle Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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