

# Graphing Periodic Trends Lab

Name \_\_\_\_\_ Hr \_\_\_\_\_

**PURPOSE:** To determine how certain properties are periodic when the elements are arranged in periods or groups.

**DISCUSSION:** There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

## REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

Ionization energy - The energy required to remove an electron from an atom.

## DATA

**Properties of the First 18 Elements**

Element	Atomic Number	Atomic Radius (Angstroms)	Ionization Energy
H	1	0.79	314
He	2	0.49	567
Li	3	2.05	124
Be	4	1.4	215
B	5	1.17	191
C	6	0.91	260
N	7	0.75	335
O	8	0.65	314
F	9	0.57	402
Ne	10	?	?
Na	11	2.23	119
Mg	12	1.72	176
Al	13	1.82	138
Si	14	1.46	168
P	15	1.23	242
S	16	1.09	239
Cl	17	0.97	299
Ar	18	?	?

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with question marks.

1. Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)
2. Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements – 3 graphs)

# Periodic Trends Graphing Lab Answers

**Clifford Lane**



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*Computer Based Projects for a Chemistry Curriculum* Thomas J. Manning, Aurora P. Gramatges, 2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom

**General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems

**Laboratory Manual** Jo A. Beran, 1990 **The Science Teachers Bulletin**, 1989

**Chemistry** Karen Timberlake, 1999 Suitable for one or two term lab courses covering general organic and biological chemistry this new edition written by Karen Timberlake features many improvements to the insightful experiments that have made it the leading lab manual Each experiment encourages critical thinking with laboratory goals discussion of related concepts clear instructions new pre lab questions and comprehensive report pages Forty one experiments illustrate the basic principles of chemistry

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## **Table of Contents Periodic Trends Graphing Lab Answers**

1. Understanding the eBook Periodic Trends Graphing Lab Answers
  - The Rise of Digital Reading Periodic Trends Graphing Lab Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Graphing Lab Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Trends Graphing Lab Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Graphing Lab Answers
  - Personalized Recommendations
  - Periodic Trends Graphing Lab Answers User Reviews and Ratings
  - Periodic Trends Graphing Lab Answers and Bestseller Lists
5. Accessing Periodic Trends Graphing Lab Answers Free and Paid eBooks
  - Periodic Trends Graphing Lab Answers Public Domain eBooks
  - Periodic Trends Graphing Lab Answers eBook Subscription Services
  - Periodic Trends Graphing Lab Answers Budget-Friendly Options
6. Navigating Periodic Trends Graphing Lab Answers eBook Formats

- ePub, PDF, MOBI, and More
- Periodic Trends Graphing Lab Answers Compatibility with Devices
- Periodic Trends Graphing Lab Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Trends Graphing Lab Answers
  - Highlighting and Note-Taking Periodic Trends Graphing Lab Answers
  - Interactive Elements Periodic Trends Graphing Lab Answers
- 8. Staying Engaged with Periodic Trends Graphing Lab Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Trends Graphing Lab Answers
- 9. Balancing eBooks and Physical Books Periodic Trends Graphing Lab Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Trends Graphing Lab Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Trends Graphing Lab Answers
  - Setting Reading Goals Periodic Trends Graphing Lab Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Trends Graphing Lab Answers
  - Fact-Checking eBook Content of Periodic Trends Graphing Lab Answers
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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