

The Periodic Table

The Structure of the Periodic Table

- The periodic table displays the elements in order of increasing atomic number i.e. in order of increasing number of protons.
- Elements are arranged in 18 vertical columns called groups that are numbered 1 – 18.
- The seven horizontal rows in periodic table are called periods. The periods are numbered 1-7. The first period contains only two elements, hydrogen and helium.
- Elements within the same groups have similar chemical properties because they have the same number of valence electrons. The number of valence electrons is the same as the group number (in Roman numerals).

<div>PERIODS</div> <div>1-7</div>		GROUPS 1-18																				
																		17	18			
		1																	17	18		
		2	Li	Be													B	C	N	O	F	Ne
		3	Na	Mg													Al	Si	P	S	Cl	Ar
		4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr		
		5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe		
		6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
		7	Fr	Ra	Ac	Rf	Ha	Sg	Nh	Hs	Mt	Ds	Ds									

Energy levels and period numbers

The electrons of an atom are arranged in energy levels. The period number tells us in which energy level an atom has its valence electrons.

For example:

Element	In which Group?	In which Period?	How many valence electrons	Which energy level are valence electrons?
Phosphorous	15	3	5	3 rd energy level
Potassium				
Calcium				
Oxygen				

Periodic Table Lessons Fourth Grade

Victor M. Corman



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Lesson Design for Differentiated Instruction, Grades 4-9 Kathy Tuchman Glass, 2009-01-14 This user friendly resource provides step by step guidance and a detailed template for creating meaningful lessons that are differentiated according to students learning characteristics

Arun Deep's Self-Help to ISC Chemistry Class 11 : For 2025-26 Examinations Amar Nath Bhutani, Saurabh Joshi, 2025-07-07 Arun Deep's Self Help to ISC Chemistry Class 11 For 2025 26 Examinations This guidebook has been meticulously crafted to support students of Class 11 who are preparing for the ISC Chemistry examination for the academic year 2025 26 Aligned with the latest ISC curriculum the book provides comprehensive solutions and explanations to all the questions presented in the ISC Chemistry textbook published by Nageen Prakashan The content is structured to aid conceptual clarity reinforce theoretical understanding and strengthen problem solving skills Each chapter includes Detailed answers to all in text and end of chapter questions Step by step solutions for numerical problems Additional tips and key points for effective revision Supportive content that complements classroom learning An ideal companion for ISC students this Self Help book aims to simplify complex concepts and provide exam oriented preparation helping learners achieve academic excellence with confidence

ISC Chemistry Book 1 for Class XI (2021 Edition) R.D.MADAN, ISC Chemistry Book 1

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Conceptual Chemistry Class XI Vol. I S K Jain, A book on Conceptual Chemistry

Self-Help to I.C.S.E.

Chemistry Class 10 (For 2022-23 Examinations) Amar Bhutani, RP Gupta K., This book is written strictly in accordance with the latest syllabus prescribed by the Council for the I C S E Examinations in and after 2023 This book includes the Answers to the Questions given in the Textbook Concise Chemistry Class 10 published by Selina Publications Pvt Ltd This book is written by Sunil Manchanda

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Oswal - Gurukul Chemistry Most Likely Question Bank : ICSE Class 10 For 2023 Exam Oswal - Gurukul, 2022-05-14

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CBSE New Pattern Chemistry Class 11 for 2021-22 Exam (MCQs based book for Term 1) Shahana Ansari, Abhishika Kaushik, 2021-09-10

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