

Name: _____

Dates:

CHEMISTRY

Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The periodic table is a tabular arrangement of the chemical elements, organized on the basis of their atomic number, electron configuration, and recurring chemical properties. The elements are color-coded by groups: alkali metals (orange), alkaline earth metals (yellow), transition metals (green), lanthanides (teal), noble gases (blue), halogens (purple), and metalloids (light green). The table includes element symbols, names, atomic numbers, and atomic weights. A legend at the top right explains the color coding. The table is divided into blocks: s-block, p-block, d-block, and f-block.

Part 1: Atomic Number. On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- | | | | | | |
|----|--|-----------|-----|--|--------------|
| 1. | | Hydrogen | 7. | | Cerium |
| 2. | | Oxygen | 8. | | Iridium |
| 3. | | Magnesium | 9. | | Bismuth |
| 4. | | Vanadium | 10. | | Uranium |
| 5. | | Krypton | 11. | | Fermium |
| 6. | | Ruthenium | 12. | | Darmstadtium |

Periodic Table Answers Section 3

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