

Experiment 11 Prelaboratory Assignment

Periodic Table and Periodic Law

Date 2/3/24 Lab Sec. _____ Name [REDACTED] Desk No. _____

1. On the blank periodic table, clearly indicate six of the following element sections, using your own color code.
- a. Representative elements
 - b. Transition elements
 - c. Inner-transition elements
 - d. Chalcogens ✱
 - e. Coinage metals
 - f. Metalloids
 - g. Post-transition metals
 - ✓ h. Alkali metals
 - ✓ i. Alkaline earth metals
 - ✓ j. Halogens ✱
 - ✓ k. Noble gases
 - l. Noble metals
 - ✓ m. Lanthanide series
 - ✓ n. Actinide series

Periodic Table

2. Sketch in the staircase line that separates the metals from the nonmetals on the periodic table.
in black
3. Identify the atomic numbers of the elements that would be called the *transactinide* elements.

atomic # 104 - 112
(greater than 103)

Periodic Law Lab Unscrambled Table Answers

Eric R. Scerri



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