



ACS  
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# PERIODIC TABLE OF ELEMENTS

group	1	2											13	14	15	16	17	18
1	<div>H</div> <div>Hydrogen</div> <div>1</div> <div>1.008</div>																	<div>He</div> <div>Helium</div> <div>2</div> <div>4.0026</div>
2	<div>Li</div> <div>Lithium</div> <div>3</div> <div>6.941</div>	<div>Be</div> <div>Beryllium</div> <div>4</div> <div>9.0122</div>											<div>B</div> <div>Boron</div> <div>5</div> <div>10.811</div>	<div>C</div> <div>Carbon</div> <div>6</div> <div>12.011</div>	<div>N</div> <div>Nitrogen</div> <div>7</div> <div>14.0064</div>	<div>O</div> <div>Oxygen</div> <div>8</div> <div>15.9994</div>	<div>F</div> <div>Fluorine</div> <div>9</div> <div>18.9984</div>	<div>Ne</div> <div>Neon</div> <div>10</div> <div>20.1798</div>
3	<div>Na</div> <div>Sodium</div> <div>11</div> <div>22.9898</div>	<div>Mg</div> <div>Magnesium</div> <div>12</div> <div>24.3047</div>											<div>Al</div> <div>Aluminum</div> <div>13</div> <div>26.9815</div>	<div>Si</div> <div>Silicon</div> <div>14</div> <div>28.0855</div>	<div>P</div> <div>Phosphorus</div> <div>15</div> <div>30.9738</div>	<div>S</div> <div>Sulfur</div> <div>16</div> <div>32.065</div>	<div>Cl</div> <div>Chlorine</div> <div>17</div> <div>35.453</div>	<div>Ar</div> <div>Argon</div> <div>18</div> <div>39.948</div>
4	<div>K</div> <div>Potassium</div> <div>19</div> <div>39.0983</div>	<div>Ca</div> <div>Calcium</div> <div>20</div> <div>40.078</div>	<div>Sc</div> <div>Scandium</div> <div>21</div> <div>44.9559</div>	<div>Ti</div> <div>Titanium</div> <div>22</div> <div>47.88</div>	<div>V</div> <div>Vanadium</div> <div>23</div> <div>50.9415</div>	<div>Cr</div> <div>Chromium</div> <div>24</div> <div>51.9961</div>	<div>Mn</div> <div>Manganese</div> <div>25</div> <div>54.938</div>	<div>Fe</div> <div>Iron</div> <div>26</div> <div>55.845</div>	<div>Co</div> <div>Cobalt</div> <div>27</div> <div>58.9332</div>	<div>Ni</div> <div>Nickel</div> <div>28</div> <div>58.6934</div>	<div>Cu</div> <div>Copper</div> <div>29</div> <div>63.546</div>	<div>Zn</div> <div>Zinc</div> <div>30</div> <div>65.38</div>	<div>Ga</div> <div>Gallium</div> <div>31</div> <div>69.723</div>	<div>Ge</div> <div>Germanium</div> <div>32</div> <div>72.64</div>	<div>As</div> <div>Arsenic</div> <div>33</div> <div>74.9216</div>	<div>Se</div> <div>Selenium</div> <div>34</div> <div>78.96</div>	<div>Br</div> <div>Bromine</div> <div>35</div> <div>79.904</div>	<div>Kr</div> <div>Krypton</div> <div>36</div> <div>83.798</div>
5	<div>Rb</div> <div>Rubidium</div> <div>37</div> <div>85.4678</div>	<div>Sr</div> <div>Strontium</div> <div>38</div> <div>87.62</div>	<div>Y</div> <div>Yttrium</div> <div>39</div> <div>88.9058</div>	<div>Zr</div> <div>Zirconium</div> <div>40</div> <div>91.224</div>	<div>Nb</div> <div>Niobium</div> <div>41</div> <div>92.9064</div>	<div>Mo</div> <div>Molybdenum</div> <div>42</div> <div>95.94</div>	<div>Tc</div> <div>Technetium</div> <div>43</div> <div>98</div>	<div>Ru</div> <div>Ruthenium</div> <div>44</div> <div>101.07</div>	<div>Rh</div> <div>Rhodium</div> <div>45</div> <div>102.91</div>	<div>Pd</div> <div>Palladium</div> <div>46</div> <div>106.42</div>	<div>Ag</div> <div>Silver</div> <div>47</div> <div>107.87</div>	<div>Cd</div> <div>Cadmium</div> <div>48</div> <div>112.41</div>	<div>In</div> <div>Indium</div> <div>49</div> <div>114.82</div>	<div>Sn</div> <div>Tin</div> <div>50</div> <div>118.71</div>	<div>Sb</div> <div>Antimony</div> <div>51</div> <div>121.76</div>	<div>Te</div> <div>Tellurium</div> <div>52</div> <div>127.6</div>	<div>I</div> <div>Iodine</div> <div>53</div> <div>126.91</div>	<div>Xe</div> <div>Xenon</div> <div>54</div> <div>131.29</div>
6	<div>Cs</div> <div>Cesium</div> <div>55</div> <div>132.91</div>	<div>Ba</div> <div>Barium</div> <div>56</div> <div>137.33</div>	<div>La</div> <div>Lanthanum</div> <div>57</div> <div>138.91</div>	<div>Hf</div> <div>Hafnium</div> <div>58</div> <div>178.49</div>	<div>Ta</div> <div>Tantalum</div> <div>59</div> <div>180.95</div>	<div>W</div> <div>Tungsten</div> <div>60</div> <div>183.84</div>	<div>Re</div> <div>Rhenium</div> <div>61</div> <div>186.21</div>	<div>Os</div> <div>Osmium</div> <div>62</div> <div>190.23</div>	<div>Ir</div> <div>Iridium</div> <div>63</div> <div>192.22</div>	<div>Pt</div> <div>Platinum</div> <div>64</div> <div>195.08</div>	<div>Au</div> <div>Gold</div> <div>65</div> <div>196.97</div>	<div>Hg</div> <div>Mercury</div> <div>66</div> <div>200.59</div>	<div>Tl</div> <div>Thallium</div> <div>67</div> <div>204.38</div>	<div>Pb</div> <div>Lead</div> <div>68</div> <div>207.2</div>	<div>Bi</div> <div>Bismuth</div> <div>69</div> <div>208.98</div>	<div>Po</div> <div>Polonium</div> <div>70</div> <div>209</div>	<div>At</div> <div>Astatine</div> <div>71</div> <div>210</div>	<div>Rn</div> <div>Radon</div> <div>72</div> <div>222</div>
7	<div>Fr</div> <div>Francium</div> <div>73</div> <div>223</div>	<div>Ra</div> <div>Radium</div> <div>74</div> <div>226</div>	<div>Ac</div> <div>Actinium</div> <div>75</div> <div>227</div>	<div>Rf</div> <div>Rutherfordium</div> <div>76</div> <div>261</div>	<div>Db</div> <div>Dubnium</div> <div>77</div> <div>262</div>	<div>Sg</div> <div>Seaborgium</div> <div>78</div> <div>266</div>	<div>Bh</div> <div>Bohrium</div> <div>79</div> <div>264</div>	<div>Hs</div> <div>Hassium</div> <div>80</div> <div>277</div>	<div>Mt</div> <div>Meitnerium</div> <div>81</div> <div>268</div>	<div>Ds</div> <div>Darmstadtium</div> <div>82</div> <div>271</div>	<div>Rg</div> <div>Roentgenium</div> <div>83</div> <div>272</div>	<div>Cn</div> <div>Copernicium</div> <div>84</div> <div>285</div>	<div>Nh</div> <div>Nihonium</div> <div>85</div> <div>284</div>	<div>Fl</div> <div>Flerovium</div> <div>86</div> <div>289</div>	<div>Mc</div> <div>Moscovium</div> <div>87</div> <div>288</div>	<div>Lv</div> <div>Livermorium</div> <div>88</div> <div>293</div>	<div>Ts</div> <div>Tennessine</div> <div>89</div> <div>289</div>	<div>Og</div> <div>Oganesson</div> <div>90</div> <div>294</div>
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# Period Table S Most Wanted Answers

**Julia Schneider**



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