

Salters OCR B chemistry EL not mine

2 bond pairs, 0 lone pairs - ANSLinear

180°

e.g. BeCl₂

3 bond pairs, 0 lone pairs - ANSTrigonal Planar

120°

e.g. BF₃

2 bond pairs, 1 lone pair - ANSBent

118°

e.g. NO₂

4 bond pairs, 0 lone pairs - ANSTetrahedral

109.5°

e.g. CH₄

3 bond pairs, 1 lone pair - ANSPyramid

107°

e.g. NH₃

2 bond pairs, 2 lone pairs - ANSBent

104.5°

e.g. H₂O

5 bond pairs, 0 lone pairs - ANSTrigonal Bipyramid

120°, 90°

e.g. PCl₅

6 bond pairs, 0 lone pairs - ANSOctahedral

90°

e.g. SF₆

In 1803 John Dalton said... - ANSThat all atoms are SPHERES and all elements are made from different spheres

Plum pudding model - ANSIn 1897 J.J. Thompson discovered the electron and said there were negative electrons "plums" in a positive "pudding"

Gold Leaf Experiment - ANS1909, Ernest Rutherford.

Fired positive alpha particles at gold leaf. Most went through to other side, some deflected back.

He concluded that the atom had a very small positively charged nucleus and a negative cloud of empty space.

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M Woodhall



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