

Note:

- i. Use the same answer-sheet for Section A and Section B.
- ii. Draw well-labelled diagrams wherever necessary.
- iii. All questions are compulsory.
- iv. Students should write the answers of questions in sequence.

SECTION A

1. (A) Fill in the blanks: [2]

1. The chemical formula of rust is _____
2. The law used by Newlands to arrange elements is known as _____

- (B) State whether the following statements are True or False: [2]

1. The pH of rain water is 7.
2. The SI unit of charge is volt.

- (C) Rewrite the following table so as to match the second and third columns with the first column: [3]

	Column I	Column II	Column III
1.	Dispersion	Long-sightedness	Twinkling of stars
2.	Refraction	Splitting of white light into component colours	Convex lens
3.	Hypermetropia	Change in the direction of ray of light due to change in medium	Spectrum of seven colours

2. Answer the following questions: [6]

1. Distinguish between Conductors and Insulators.
2. Why should the wires carrying electricity not be touched bare-footed?
3. Calculate the Focal length of a corrective lens having power + 2.5 D.

3. Answer the following questions (any four): [12]

1. The atomic masses of three elements A, B and C having similar chemical properties are 7, 23 and 39 respectively.
 - a. Calculate the average atomic mass of elements A and C.
 - b. Compare the average atomic mass with atomic mass of B.
 - c. What could the elements A, B and C be?
2. When the substance 'A' is added to a solution of BaSO_4 a yellow precipitate is formed.
 - a. What do you think substance A is likely to be?
 - b. Name the precipitate.
 - c. Which type of reaction is this?
3. A person has sour taste in mouth and heart burn.
 - a. What is he suffering from?
 - b. Why does it happen?
 - c. What substance is used as remedy?

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