# CHEM 101 Final Exam with complete solutions

How many significant figures are in this number, 0.01160 - ANSWER: → 4

Select the correct scientific notation for 117000 - ANSWER:  $\rightarrow 1.17 \times 10^5$ 

the temperature of an experiment is designed at 122 degrees F. What is it in Celsius? - ANSWER: → 50 degrees Celsius

An experiment is performed at -173 degrees C. What is the temperature in Kelvin? - ANSWER:  $\rightarrow$  100K

A piece of metal has a mass of 340 g and a volume of 17.6 mL. What is the density of the metal? - ANSWER: → 19.3 g/mL

The comparison of the density of a substance with the density of water is defined as \_\_\_\_\_, - ANSWER: → specific gravity

The following forms of energy can be classified as potential energy EXCEPT \_\_\_\_\_.

- A). Gasoline B). Running
- C). Peanut butter sandwich D). Water held by a dam ANSWER: running

The following examples are chemical changes EXCEPT

- A). Rusting iron B). Cooking an egg
- C). Burning gasoline D). Melting ice ANSWER: → melting ice

The amount of heat necessary to raise the temperature of 1 g liquid water by 1oC, is

defined as \_\_\_. - ANSWER: → a calorie

# **Possible Objective Questions On Chem 101 With Solutions**

**Arihant Experts** 

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helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam

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