#### Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion	Polyatomic Anion Name
HCIO.	Chloric acid	H,O	C/O3-1	Chlorate
H.SO,	Sulfurous acid	24,00	503-2	Sulfire
H.SO.	Sulfuric acid	2 Hso*	504-3	Sulfare
H,PO,	Phosphorous acid	3 H,O	P03-3	Phosphite
H.PO.	Phosphoric acid	3 H50+	PO <sub>a</sub> s-	Phosphate
HNO,	Nitric acid	H30+	NO."	Nitrose
HNO,	Nitrous acid	H.0*	NO,	Nitrite
H,CO,	Carbonic acid	2450+	CO,-2	Carbonate

- 6. Look at the formulas of the ternary acids in Model 2.
  - w. How are ternary acids different from binary acids in their structure?

Ternory acids contain three elements and they also contain the element oxygen

6. What number do you think the prefix "ter-" refers to?

"ter" rofers to three

- 7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.
- 8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is /C

Polyatomic anion ending is "-ite" → acid name ending is 045 ...

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when

- naming HCIO? CIO3' has the name of Chlorate. "ate" must be replaced with "ic" If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula HCI, not HCIO3.

  The prefix "hydro" is not used in the name when an acid contain more than two elements and one of the elements is oxygen.

  "ate is replicate with it and "ite" is replicate with "ous".

Circle the acid(s) below that would be named beginning with the prefix "hydro-."

FESO.

H,CO.

HNO.

# **Polyatomic Ions Worksheet Pogil**

Katrin Zwirglmaier

**Polyatomic Ions Worksheet Pogil:** 

The book delves into Polyatomic Ions Worksheet Pogil. Polyatomic Ions Worksheet Pogil is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and indepth insights into Polyatomic Ions Worksheet Pogil, encompassing both the fundamentals and more intricate discussions.

- 1. The book is structured into several chapters, namely:
  - Chapter 1: Introduction to Polyatomic Ions Worksheet Pogil
  - Chapter 2: Essential Elements of Polyatomic Ions Worksheet Pogil
  - Chapter 3: Polyatomic Ions Worksheet Pogil in Everyday Life
  - Chapter 4: Polyatomic Ions Worksheet Pogil in Specific Contexts
  - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Polyatomic Ions Worksheet Pogil. The first chapter will explore what Polyatomic Ions Worksheet Pogil is, why Polyatomic Ions Worksheet Pogil is vital, and how to effectively learn about Polyatomic Ions Worksheet Pogil.
- 3. In chapter 2, the author will delve into the foundational concepts of Polyatomic Ions Worksheet Pogil. The second chapter will elucidate the essential principles that must be understood to grasp Polyatomic Ions Worksheet Pogil in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Polyatomic Ions Worksheet Pogil in daily life. This chapter will showcase real-world examples of how Polyatomic Ions Worksheet Pogil can be effectively utilized in everyday scenarios.
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- 6. In chapter 5, this book will draw a conclusion about Polyatomic Ions Worksheet Pogil. The final chapter will summarize the key points that have been discussed throughout the book.
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