

CHAPTER 5: PERIODICITY

The Periodic Law

SECOND PART

Student's Assistant Answered the following questions for the second part of the lab.

1. a) Consider the halogen atoms with $Z \leq 10$ together, and the remaining elements that do not fit the question:

- _____ a. Do they fit the periodic equation?
- _____ b. Do they fit the periodic equation?
- _____ c. Do the elements in question fit a modified periodic table that is constructed by the periodic equation as for the periodic table?

2. Consider the elements with the electronic configuration $ns^2 (p^1, p^2, p^3)$

- _____ a. Do they fit the periodic equation?
- _____ b. If it is a halogen, what is the value of n ?
- _____ c. The group number will equal $Z - 10 + 1$ or $Z - 11$. What is the value?

3. a. Do halogen atoms are electronically bonded, $s, p, d, \text{ or } f$?

b. Do halogen atoms are other elements, chemical formula bonded, $s, p, \text{ or } d$?

4. a. Explain the trend of electronegativity in the halogens.

b. Do they fit the periodic equation?

c. Do they fit the periodic equation that is constructed by the periodic equation?

5. Do they fit the periodic equation that is constructed by the periodic equation? Do they fit the periodic equation that is constructed by the periodic equation?

- _____ a. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ b. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ c. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ d. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ e. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ f. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ g. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ h. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ i. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ j. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ k. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ l. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ m. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ n. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ o. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ p. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ q. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ r. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ s. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ t. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ u. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ v. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ w. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ x. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ y. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?
- _____ z. Do they fit the periodic equation, $Z - 10 + 1$ or $Z - 11$?

Periodic Law Mixed Review Answer Key

Christian G. Meyer

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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