

Periodic Trends Worksheet

Using a few minutes before the beginning activity, complete the following questions. Please post your answers on the board when you are done.

Additional help on the periodic trends can be found on page 104 of the student textbook. <https://www.ck12.org/c/chemistry/periodic-table-of-elements-2/learn/section-104/chemistry/>

1. What does electronegativity measure and how does it change across the periodic table? (electronegativity is largest) in which direction, left-to-right, up-to-down, and bottom-to-top?
2. Electronegativity values increase from left to right and bottom to top. How does this affect the periodic table?
3. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
4. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
5. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
6. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
7. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
8. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
9. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O
10. Electronegativity increases across the periodic table from left to right and bottom to top. How does this affect the periodic table?
 - a. Fluorine is F
 - b. Chlorine is Cl
 - c. Oxygen is O

Periodic Trends Questions And Answers

Albert A Gayle



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