

PERIODIC TABLE

(BASIC)

Atomic #	Symbol	Atomic Weight	State	Category
1	H	1.00794	Gas	Nonmetals
2	He	4.00260	Gas	Noble gases
3	Li	6.941	Solid	Alkali metals
4	Be	9.01218	Solid	Alkaline earth metals
5	B	10.811	Solid	Metals
6	C	12.011	Solid	Nonmetals
7	N	14.007	Gas	Nonmetals
8	O	15.999	Gas	Nonmetals
9	F	18.998	Gas	Nonmetals
10	Ne	20.180	Gas	Noble gases
11	Na	22.990	Solid	Alkali metals
12	Mg	24.305	Solid	Alkaline earth metals
13	Al	26.982	Solid	Metals
14	Si	28.086	Solid	Metals
15	P	30.974	Solid	Nonmetals
16	S	32.06	Solid	Nonmetals
17	Cl	35.45	Gas	Nonmetals
18	Ar	39.948	Gas	Noble gases
19	K	39.098	Solid	Alkali metals
20	Ca	40.078	Solid	Alkaline earth metals
21	Sc	44.956	Solid	Transition metals
22	Ti	47.88	Solid	Transition metals
23	V	50.942	Solid	Transition metals
24	Cr	51.996	Solid	Transition metals
25	Mn	54.938	Solid	Transition metals
26	Fe	55.845	Solid	Transition metals
27	Co	58.933	Solid	Transition metals
28	Ni	58.69	Solid	Transition metals
29	Cu	63.546	Solid	Transition metals
30	Zn	65.38	Solid	Transition metals
31	Ga	69.723	Solid	Post-transition metals
32	Ge	72.63	Solid	Metals
33	As	74.922	Solid	Nonmetals
34	Se	78.96	Solid	Nonmetals
35	Br	79.904	Liquid	Nonmetals
36	Kr	83.80	Gas	Noble gases
37	Rb	85.468	Solid	Alkali metals
38	Sr	87.62	Solid	Alkaline earth metals
39	Y	88.906	Solid	Transition metals
40	Zr	91.224	Solid	Transition metals
41	Nb	92.906	Solid	Transition metals
42	Mo	95.94	Solid	Transition metals
43	Tc	98.906	Solid	Transition metals
44	Ru	101.07	Solid	Transition metals
45	Rh	102.91	Solid	Transition metals
46	Pd	106.42	Solid	Transition metals
47	Ag	107.87	Solid	Transition metals
48	Cd	112.41	Solid	Transition metals
49	In	114.82	Solid	Post-transition metals
50	Sn	118.71	Solid	Metals
51	Sb	121.76	Solid	Metals
52	Te	127.6	Solid	Nonmetals
53	I	126.90	Solid	Nonmetals
54	Xe	131.29	Gas	Noble gases
55	Ba	137.33	Solid	Alkaline earth metals
56	La	138.90	Solid	Transition metals
57	Ce	140.12	Solid	Transition metals
58	Pr	140.91	Solid	Transition metals
59	Nd	144.24	Solid	Transition metals
60	Pm	144.91	Solid	Transition metals
61	Sm	150.36	Solid	Transition metals
62	Eu	151.96	Solid	Transition metals
63	Gd	157.25	Solid	Transition metals
64	Tb	158.93	Solid	Transition metals
65	Dy	162.50	Solid	Transition metals
66	Ho	164.93	Solid	Transition metals
67	Er	167.26	Solid	Transition metals
68	Tm	168.93	Solid	Transition metals
69	Yb	173.05	Solid	Transition metals
70	Lu	174.96	Solid	Transition metals
71	Hf	178.49	Solid	Transition metals
72	Ta	180.95	Solid	Transition metals
73	W	183.84	Solid	Transition metals
74	Re	186.21	Solid	Transition metals
75	Os	190.23	Solid	Transition metals
76	Ir	192.22	Solid	Transition metals
77	Pt	195.08	Solid	Transition metals
78	Au	196.97	Solid	Transition metals
79	Hg	200.59	Liquid	Transition metals
80	Tl	204.38	Solid	Post-transition metals
81	Pb	207.2	Solid	Metals
82	Bi	208.98	Solid	Metals
83	Po	209	Solid	Metals
84	At	210	Solid	Metals
85	As	210	Solid	Metals
86	Rn	222	Gas	Noble gases
87	Fr	223	Solid	Alkali metals
88	Ra	226	Solid	Alkaline earth metals
89	Ac	227	Solid	Transition metals
90	Th	232.04	Solid	Transition metals
91	Pa	231.04	Solid	Transition metals
92	U	238.03	Solid	Transition metals
93	Np	237.05	Solid	Transition metals
94	Pu	244.06	Solid	Transition metals
95	Am	243.06	Solid	Transition metals
96	Cm	247.07	Solid	Transition metals
97	Bk	247.07	Solid	Transition metals
98	Cf	251.08	Solid	Transition metals
99	Es	252.08	Solid	Transition metals
100	Fm	257.10	Solid	Transition metals
101	Md	258.10	Solid	Transition metals
102	No	259.10	Solid	Transition metals
103	Lr	262.11	Solid	Transition metals

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

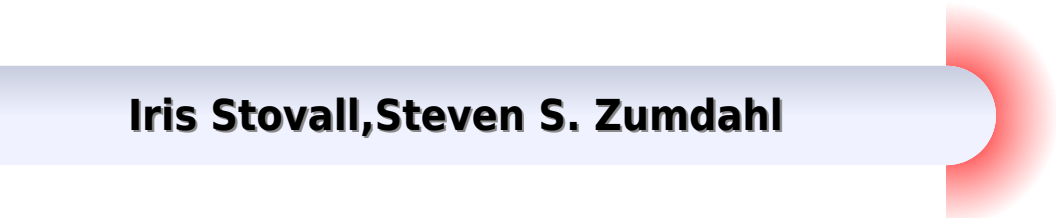
Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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Iris Stovall, Steven S. Zumdahl



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Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry Iris Stovall,Steven S. Zumdahl,1996

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