

2-2

Possible stereoisomers are:

a) **Eis**: **Tiis**, mixture (because of double bond)
 b) isotactic, syndiotactic, atactic (because of -CHCl- groups). However, because of distance between -CHCl- groups tacticity is unlikely to influence ability to crystallize unless rest of groups are all gls or trans.

Use $\begin{bmatrix} 1 \\ 1.51 \end{bmatrix}$ for $\begin{matrix} \text{C-O} = a \\ \text{O-C} = b \end{matrix}$ $\begin{matrix} 108^\circ \text{ for C-O-C} = \theta \\ 110^\circ \text{ for C-C-O} = \theta \end{matrix}$

2-3

Graphical solution is easiest using protractor and ruler. Vector summation can be used, also.



If by syndiotactic we mean a succession of **d** and **l** monomer units, the distance turns out to be the same. Another structure is possible with methyls alternating so that the repeat distance would be doubled.

2-4

Refer to Table 2-1, Fig. 2-1.
 Polycyclohexane in cyclohexane:

$$X = 0.136, \quad V_u = 108, \quad v_2 = 0.10$$

$$NV = 2.45 \times 10^3, \quad \text{so that } N = 2.27 \times 10^5 \text{ units/cm}^3$$

Same in solution:

$$X = 0.557, \quad V_u = 114, \quad W_u = 2.40 \times 10^{-22}, \quad \text{then } v_2 = 0.21$$

$\frac{10^{-22} \text{ moles}}{1 \text{ g.}} \div \frac{1}{\text{original volume}}$

2-5

$$N = 0.5, \quad \text{moles/volume} = (\text{g/vol})/(\text{g/mole})$$

If $M = 161.91$, per repeat unit = 38 for $\text{C}_6\text{H}_6\text{O}$ C chain atoms)

then $M_c = 161.91$, between cross-links = $\frac{5000 \times 38}{5}$ g/mole

$$\rho = 5313, \quad a = 10^{-8} \text{ meter}$$

$$V = 102 \times 1.21 \times 10^{-22} / 0.800 = 1.58 \times 10^{-22}$$

From Fig. 2-1: $v_2 = 0.059$

same additive volumes. Basis: Total volume = 1 cm³

$$1.0 \text{ c7} = V + V' + V_2 = 0.931 + 0.069$$

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