Section 1.4: Alcohols, Ethers, and Thiols

Tutorial 1 Practice, page 34

(a) Step 1: In this compound, the longest chain has 5 carbon atoms: pentane.

Step 2: There are two -OH groups, so the compound is a pentanediol.

Step 3: The -OH groups are attached to carbons 1 and 4.

Step 4: This compound is pentane-1,4-diol.

(b) Step 1: In this compound, the longest chain has 8 carbon atoms: octane.

Step 2: There is one -OH group, so the compound is an octanol.

Step 3: The –OH group is attached to carbon 4, using the lowest possible number, so the compound is an octan-4-ol.

Step 4: There is a methyl group attached carbon 3 and an ethyl group attached to carbon 4. This compound is 4-ethyl-3-methyloctan-4-ol.

(c) Step 1: In this compound, the parent structure is benzene.

Step 2: There are two -OH groups, so the compound is a benzenediol.

Step 3: The -OH groups are attached to carbons 1 and 3, so the compound is benzene-1,3-diol.

2. (a) The name "heptanol" indicates that this alcohol consists of a 7-carbon chain with one hydroxyl group attached to it. The hydroxyl group is attached to carbon 3. Add the substituents as indicated in the name: a chlorine atom on carbon 2 and two methyl groups on carbons 2 and 5. Therefore, the structure of 2-chloro-2,5-dimethylheptan-3-ol is

(b) The name "propanediol" indicates that this alcohol consists of a 3-carbon chain with two hydroxyl groups attached to it. The hydroxyl groups are attached to carbons 1 and 3. Therefore, the structure of propane-1,3-diol is

(c) The simplest aromatic alcohol is a benzene ring with one hydroxyl group bonded to it. The IUPAC name of this compound is phenol. Therefore, the structure of phenol is

(d) The name "pentenediol" indicates that this alcohol consists of a 5-carbon chain with a double bond and two hydroxyl groups attached to it. The hydroxyl groups are attached to carbons 1 and 2, and the double bond is between carbons 2 and 3. Therefore, the structure of pent-2-ene-1,2-diol is:

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- 1. Understanding the eBook Nelson Chemistry 12 Solutions
 - The Rise of Digital Reading Nelson Chemistry 12 Solutions
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Nelson Chemistry 12 Solutions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nelson Chemistry 12 Solutions
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Nelson Chemistry 12 Solutions
 - Personalized Recommendations
 - Nelson Chemistry 12 Solutions User Reviews and Ratings
 - Nelson Chemistry 12 Solutions and Bestseller Lists
- 5. Accessing Nelson Chemistry 12 Solutions Free and Paid eBooks

- Nelson Chemistry 12 Solutions Public Domain eBooks
- Nelson Chemistry 12 Solutions eBook Subscription Services
- Nelson Chemistry 12 Solutions Budget-Friendly Options
- 6. Navigating Nelson Chemistry 12 Solutions eBook Formats
 - o ePub, PDF, MOBI, and More
 - Nelson Chemistry 12 Solutions Compatibility with Devices
 - Nelson Chemistry 12 Solutions Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nelson Chemistry 12 Solutions
 - Highlighting and Note-Taking Nelson Chemistry 12 Solutions
 - Interactive Elements Nelson Chemistry 12 Solutions
- 8. Staying Engaged with Nelson Chemistry 12 Solutions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nelson Chemistry 12 Solutions
- 9. Balancing eBooks and Physical Books Nelson Chemistry 12 Solutions
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Nelson Chemistry 12 Solutions
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nelson Chemistry 12 Solutions
 - Setting Reading Goals Nelson Chemistry 12 Solutions
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nelson Chemistry 12 Solutions
 - Fact-Checking eBook Content of Nelson Chemistry 12 Solutions
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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