

Answers to questions

Chapter 1

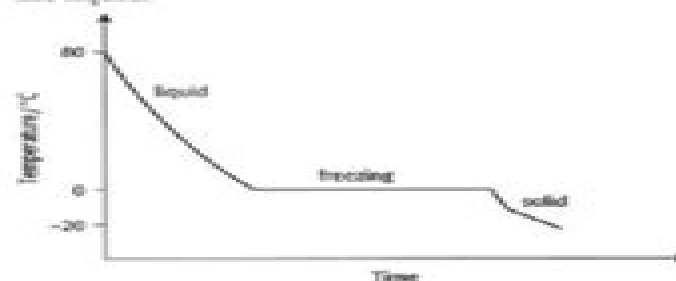
- 1.1** Petroleum and natural gas are formed from the bodies of dead marine creatures subjected to heat and pressure over very long geological periods of time.
- 1.2** It provides the energy needed for photosynthesis, which removes carbon dioxide from the atmosphere.
- 1.3** They are regarded as non-renewable because they were formed over very long periods of time and are being used up at a rate far faster than they can be formed.
- 1.4** **a** carbon dioxide + water $\rightarrow$ glucose + oxygen
(chemical equation) $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
- b** carbon + oxygen $\rightarrow$ carbon dioxide
(chemical equation) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
- c** glucose + oxygen $\rightarrow$ carbon dioxide + water
(chemical equation) $\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$
- 1.5** sulfur dioxide and nitrogen dioxide
- 1.6** the burning of fossil fuels (mainly coal for sulfur dioxide)
- 1.7** damage to limestone buildings, death of trees, acidification of lakes leading to death of fish
- 1.8** a combination of nitrogen oxides and low-level ozone that causes breathing problems, especially for people with asthma.
- 1.9** It combines with the haemoglobin in red blood cells, stopping them from carrying oxygen.
- 1.10** because it does not react with the filament, which would burn in air when it became hot
- 1.11** from rotting vegetable matter and from the intestines of animals such as cows
- 1.12** Heat which would normally escape into space is reflected back to the Earth's surface by gases such as carbon dioxide and methane in the atmosphere.
- 1.13** It changes nitrogen oxides and carbon monoxide to nitrogen and carbon dioxide.
- 1.14** because they have different boiling points
- 1.15** because solid matter is easiest to remove and would interfere with subsequent processes

- 1.16** to kill bacteria present in the water
- 1.17** because the energy needed to boil the water is costly
- 1.18** The sewage reacts with oxygen in the water, leaving less for water creatures to breathe.
- 1.19** They can cause too many algae to grow and they are not removed by water treatment.
- 1.20** It contains a high concentration of a particular metal compound.
- 1.21** Lime is calcium oxide which is reacted with water to make calcium hydroxide (slaked lime).
- 1.22** plastics, drugs, paints, detergents, etc.
- 1.23** It reflects heat back to Earth when present in the atmosphere. Heat is kept in the atmospheric layer.
- 1.24** The electrolysis of water. It is an expensive process.
- 1.25** It produces only water when burned.
- 1.26** hydrogen + oxygen $\rightarrow$ water
 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

Chapter 2

- 2.1** **a** freezing (solidification)
b boiling
c condensation
d sublimation
- 2.2** The impurity lowers the freezing point of the liquid.

2.3



- 2.4** A volatile liquid is one that evaporates easily; it has a low boiling point.

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Xiaolong Qi



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1. Understanding the eBook Nelson Chemistry 1unit Answer Key
 - The Rise of Digital Reading Nelson Chemistry 1unit Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Nelson Chemistry 1unit Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nelson Chemistry 1unit Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nelson Chemistry 1unit Answer Key
 - Personalized Recommendations

- Nelson Chemistry 1unit Answer Key User Reviews and Ratings
- Nelson Chemistry 1unit Answer Key and Bestseller Lists
- 5. Accessing Nelson Chemistry 1unit Answer Key Free and Paid eBooks
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 - Nelson Chemistry 1unit Answer Key eBook Subscription Services
 - Nelson Chemistry 1unit Answer Key Budget-Friendly Options
- 6. Navigating Nelson Chemistry 1unit Answer Key eBook Formats
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 - Nelson Chemistry 1unit Answer Key Compatibility with Devices
 - Nelson Chemistry 1unit Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nelson Chemistry 1unit Answer Key
 - Highlighting and Note-Taking Nelson Chemistry 1unit Answer Key
 - Interactive Elements Nelson Chemistry 1unit Answer Key
- 8. Staying Engaged with Nelson Chemistry 1unit Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nelson Chemistry 1unit Answer Key
- 9. Balancing eBooks and Physical Books Nelson Chemistry 1unit Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nelson Chemistry 1unit Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nelson Chemistry 1unit Answer Key
 - Setting Reading Goals Nelson Chemistry 1unit Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nelson Chemistry 1unit Answer Key
 - Fact-Checking eBook Content of Nelson Chemistry 1unit Answer Key

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- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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