- Given the nuclear equation:  $147N + X \rightarrow 162O + 21H$ What is particle X7 A) an alpha particle B) a beta particle. C) a deuteron. D) a triton.
- 2. The nucleus of a radium-226 atom is unstable, which causes the nucleus to spontaneously
  - A) absorb electrons
- B) absorb protons

C) decay

- D) oxidize
- Which group of nuclear emissions is listed in order of increasing charge?
  - A) alpha particle, beta particle, gamma radiation =
  - B) gamma radiation, alpha particle, beta particle —
  - C) positron, alpha particle, neutron
  - D) neutron, positron, alpha particle
- 4. Which list of nuclear emissions is arranged in order from the least penetrating power to the greatest penetrating p-owner?
  - A) alpha particle, beta particle, gamma ray
  - B) alpha particle, gamma ray, beta particle
  - C) gamma ray, beta particle, alpha particle
  - D) beta particle, alpha particle, gamma ray
- 5. Base your answer to the following question on Given the nuclear equation:

$$^{1}iH + X \rightarrow ^{6}iLi + ^{4}He$$

The particle represented by X is

- A) 94Li /B) 94Be C) 105Be D) 106C

- 6. Positrons are spontaneously emitted from the nuclei of
  - A) potassium-37
- B) radium-226
- C) nitrogen-16
- D) thorium-232
- 7. What is the decay mode of 37K?
  - A) 3-
- (c) (7)
- DO a
- 8. Which equation represents positron decay?
  - A) 27 77 Rb -> 0 -1 e + 27 76 Sr
  - B)  $277 \odot U \rightarrow 223 \odot Th + 40 He$
  - (c)  $^{27}_{13}Al + ^{4}_{2}He \rightarrow ^{30}_{15}P + ^{1}_{9}n$
  - D) 116 C 0+1e+115B

- 9. As <sup>14</sup>C decays to <sup>14</sup>N, the number of protons in the nucleus.
  - A) decreases

- B) increases
- C) remains the same
- 10. Atoms of I-131 spontaneously decay when the
  - A) stable nuclei emit alpha particles.
  - B) stable nuclei emit beta particles
  - C) unstable nuclei emit alpha particles
  - D) unstable nuclei emit beta particles
- 11. Which statement best describes gamma radiation?
  - A) It has a mass of 1 and a charge of 1.
  - B) It has a mass of 0 and a charge of -1.
  - (C) It has a mass of 0 and a charge of 0.
  - D) It has a mass of 4 and a charge of +2.
- 12. Which nuclear decay emission consists of energy, only?
  - A) alpha particle
- B) beta particle.
- C) gamma radiation
- D)) positron
- Which kind of nuclear radiation has high energy and no 1008.557
  - A) alpha

B) beta

C) gamma

- D) neutron
- The half-life of a radioactive isotope is 20.0 minutes. What is the total amount of a 1.00-gram sample of this isotope remaining after 1.00 hour?
  - A) 0.500 g

B) 0.333 g

C) 0.250 g

- D) 0.125 g
- 15. A radioactive element has a half-life of 2 days. Which fraction represents the amount of an original sample of this element remaining after 6 days?



# **Review Nuclear Chemistry Section 2 Answers**

**S Nieto** 

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web jun 11 2023 key concept builder lesson 2 understanding inheritance answers summarize the article in one paragraph including an overview of the topic key concepts understanding chemical reactions key concept what happens to the total

 $mass\ in\ a\ chemical\ lesson\ 1\ understanding\ chemical\ reactions\ key\ concept\ builder$