

CHAPTER 17

RADIOACTIVITY AND NUCLEAR CHEMISTRY

Practice Problems

1. (a) Write a nuclear equation for alpha decay of Po-210 .



- (b) Write a nuclear equation for the beta decay of Th-232 .



- (c) Write a nuclear equation for positron emission by Co-59 .



2. A 10.0-g sample of a radioactive nuclide is administered to a patient for cancer diagnosis. If the nuclide has a half-life of 1.5 hours, how much of the nuclide remains in the patient after 4.5 days?

$$\text{4.5 days} \times \frac{24 \text{ hours}}{1 \text{ day}} = 108 \text{ hr} \times \frac{1 \text{ day}}{24 \text{ hr}} = 4.5 \text{ days} \quad \text{Given}$$

- (a) $\text{4.5 days} \times \frac{24 \text{ hr}}{1 \text{ day}} = 108 \text{ hr}$ (b) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (c) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (d) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (e) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (f) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (g) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (h) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (i) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (j) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (k) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (l) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (m) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (n) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (o) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (p) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (q) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (r) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (s) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (t) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (u) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (v) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (w) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (x) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (y) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$ (z) $\text{4.5 days} \times \frac{1 \text{ day}}{24 \text{ hr}} = 0.1875 \text{ days}$

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- The age of the mammoth is equivalent to three half-lives of C-14 , which approximately

$$3 \times 5730 \text{ yrs} = 17,190 \text{ yrs old}$$

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