

Chapter 2 Review Sheet

1. Define chemical element, and list the four elements that form the bulk of body matter.
An element is made of just one kind of atom.
Hydrogen, Oxygen Nitrogen, Carbon (HONC)
2. List the subatomic particles, and describe their relative charges and positions in the atom. Proton-positive charge; Neutron-neutral; electron-negative charge. Protons and neutrons comprise the nucleus. Electrons orbit around the nucleus of the atom.
3. Recognize that chemical reactions involve the interaction of electrons to make and break chemical bonds.
4. Describe covalent, ionic and hydrogen bonds.
Covalent bonds are when atoms share electrons in their valence or outer shell. Ionic bonds are when atoms transfer electrons from their valence or outer shell. Hydrogen bonds are when hydrogen atoms bond with a highly electronegative atom such as oxygen. Hydrogen bonds hold water molecules together.
5. Contrast the relative strengths of covalent, ionic and hydrogen bonds. Covalent-Very strong; Ionic- Strong, but not as strong as covalent bonds; hydrogen-very weak bond
6. Describe a chemical reaction. Be able to recognize the reactants and products. A process in which one or more substances, the reactants, are converted to one or more different substances, the products. Substances are either **chemical** elements or compounds.
7. Explain the concept of pH and the pH scale. State the pH of pure water, blood, hydrochloric acid. **pH** is a measure of how acidic/basic water is. The range goes from 0 - 14, with 7 being neutral. pHs of less than 7 indicate acidity, whereas a **pH** of greater than 7 indicates a base. **pH** is really a measure of the relative amount of free hydrogen and hydroxyl ions in the water. Water = 7; blood = 7.4; Hydrochloric acid = 1-3.
8. Describe the function of carbohydrates. Carbohydrates function as an energy source. Give an example of a monosaccharide, disaccharide and a polysaccharide.
Monosaccharide-glucose; Disaccharide-Sucrose; Polysaccharide-starch
9. Draw a monosaccharide, disaccharide and polysaccharide molecule. See ppt or worksheets.
10. Describe the function of lipids. Lipids function as a long term energy source as well as insulation and protection.
11. Identify the three main types of lipids and draw their molecular structure.
1)Triglyceride; 2)Phospholipids; 3)Steroids
12. Describe the functions of proteins. They do most of the work in cells and are required for the structure, **function**, and regulation of the body's tissues and organs. ... Enzymes carry out almost all of the thousands of chemical reactions that take place in cells.

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Introduction to General, Organic, and Biochemistry Morris Hein,Scott Pattison,Susan Arena,Leo R. Best,2014-01-15 The most comprehensive book available on the subject Introduction to General Organic and Biochemistry 11th Edition continues its tradition of fostering the development of problem solving skills featuring numerous examples and coverage of current applications Skillfully anticipating areas of difficulty and pacing the material accordingly this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career **The Review of Reviews** Albert Shaw,1897 *The Review of reviews* ,1893 **Academy, with which are Incorporated Literature and the English Review** ,1878 **American Monthly Review of Reviews** Albert Shaw,1920 **Reality of Life** Syed Khasimuddin,2017-07-07 The author was born in Hyderabad India into a lower middle class family He got his high school education from a Catholic Missionary school the best in town and went on to get a bachelors and masters degree in entomology He then got into the University of California at Riverside and earned a PhD in entomology under the supervision and guidance of Prof Paul H DeBauch a world authority in biological control Growing up in India with proper social ethical values instilled by his elders and his school gave him the background to lead a decent life in different parts of the world His professional career started after his masters and includes research and teaching in entomology in India Kenya Zimbabwe England Oxford University Jordan with the USAID and USA California where he primarily worked for the state of California as a research scientist His work in Kenya was with an international research organization UN funded He has eighteen scientific publications to his credit published in internationally known recognized scientific journals In his social life he has been deeply involved in community work and drafted to lead several institutions organizations throughout his adult life He is also the founder of an Islamic school nationally recognized and a Muslim cemetery in Sacramento California Retired in 2009 he devotes most of his time to community work This current venture is an effort to convey to the reader the down to earth meaning and the real values of life the understanding of some of the natural sciences from a laymans perspective and above all leading a content life Book Review Digest ,1926 **The Complete Idiot's Guide to Nursing Entrance Exams** Maryanne Baudo, N.P-C ; M.S.N; R.N.,Robin Kavanagh,2011-09-06 Provides preparation for all five nursing entrance exams Includes sample questions in each chapter and two full fledged practice exams *Indian National Bibliography* B. S.

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