

## Chemical bonding - answers

1. Atoms form molecules by sharing electrons and those are known as covalently bonded molecules. As same, by donating electrons to non-metals from metals the ionic compounds are forms.

a) Group the following compounds – ionic or covalent.

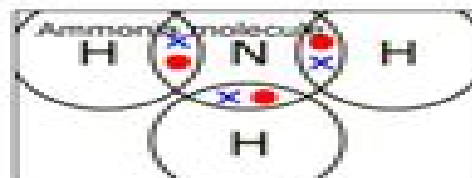
Carbon dioxide      Magnesium      Aluminium oxide      water      sulfur trioxide  
Potassium iodide      methane      ethanol      oxygen      ammonium chloride      ammonia gas  
Diamond      copper nitrate      silicon dioxide      sodium chloride      hydrogen chloride gas

| Covalent compounds | Ionic compounds  |
|--------------------|------------------|
| Carbon dioxide     | Aluminium oxide  |
| Water              | Potassium iodide |
| Sulfur trioxide    | Copper nitrate   |
| Methane            | Sodium chloride  |
| Ethanol            |                  |
| Oxygen             |                  |
| Ammonia gas        |                  |
| Silicon dioxide    |                  |
| Hydrogen gas       |                  |
|                    |                  |

b). Simple molecular compounds can be formed by sharing electrons only. Give 3 examples for simple molecular compounds.

1.....methanol ..... 2.....sulfur dioxide ..... 3.....hydrogen sulfide gas ( any covalent compound)

c. Draw dot cross diagrams to show the bonding in: i) hydrogen chloride molecule      ii) ammonia molecule



d) Explain why the bond in a nitrogen molecule is called a triple bond.

Two nitrogen atoms share three electrons from each atom to form three covalent bonds.

e). How many electrons are need to form a covalent bond?

Two electrons / a pair of electrons

# Review Chemical Bonding Answers

**Gary Raham**



## **Review Chemical Bonding Answers:**

## The Enigmatic Realm of **Review Chemical Bonding Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Review Chemical Bonding Answers** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting impact on the hearts and minds of people who partake in its reading experience.

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