



Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. ☺

1. What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_3H_8

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: $H-H$

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: $H:H$

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

3. Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl_2		<i>Note: The electron dot structures have dots in between each atom instead of lines.</i>	$:\ddot{Cl}-\ddot{Cl}:$
b) H_2S			$\begin{array}{c} :\ddot{S}: \\ \\ H \end{array} - H$
c) BrI			$:\ddot{I}-\ddot{Br}:$

Review Answers For Covalent Bonding

Marco Cascella



Review Answers For Covalent Bonding:

Whispering the Strategies of Language: An Psychological Quest through **Review Answers For Covalent Bonding**

In a digitally-driven earth where screens reign great and instant interaction drowns out the subtleties of language, the profound secrets and emotional subtleties concealed within words frequently get unheard. Yet, located within the pages of **Review Answers For Covalent Bonding** a charming literary treasure sporting with natural thoughts, lies an extraordinary journey waiting to be undertaken. Written by a talented wordsmith, that enchanting opus attracts visitors on an introspective journey, delicately unraveling the veiled truths and profound influence resonating within ab muscles cloth of each word. Within the mental depths of this touching review, we can embark upon a genuine exploration of the book is core themes, dissect their captivating publishing type, and succumb to the strong resonance it evokes strong within the recesses of readers hearts.

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