

Reversible Reactions and Chemical Equilibrium Worksheet

Supporting questions

- 1) Can you draw the reversible arrow symbol?
- 2) What do you observe when you heat up hydrated copper sulfate?
- 3) Define the term dynamic equilibrium?
- 4) Why can dynamic equilibrium only occur in a closed system?
- 5) What are the 3 factors that alter the position of equilibrium?
- 6) What happens to the position of equilibrium if you add a catalyst?
- 7) For the chemical reaction below can you say what happens to the concentration of ammonia if you (a) Add in more hydrogen (b) increase the pressure (c) increase the temperature (the forward reaction is **exothermic**).



Nitrogen Hydrogen Ammonia

- 8) Why might a high temperature still be used?

Thanks for taking the time to complete these questions, you can check your answers on the next page.

Reversible Reactions And Equilibrium Workbook Answers

Arshad Iqbal



Reversible Reactions And Equilibrium Workbook Answers:

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METALLURGICAL KINETICS GHOSH, AHINDRA, GHOSH, SUDIPTO, 2014-01-01 Mechanical kinetics constitutes one of the basic subjects for Metallurgical Engineering This well written book presents the subject of kinetics of metallurgical processes in a compressive fashion Organized into 14 chapters the book begins with an introduction of the broad basic concepts It then discusses the kinetics of homogeneous and heterogeneous chemical reactions with some real life examples from the metallurgical field The book adequately covers the concepts of diffusion convective mass transfer and mixing in fluids as well as mass transfer in fluids adjacent to a solid surface Several important processes in metallurgical and materials engineering involve reactions of porous solids with gases The book discusses this with the help of two important reactions namely reduction of iron ores and gasification of carbon It also deals with mass transfer among two fields and presents the kinetics of electrochemical reactions and phase transformation in a simple manner The book also contains plenty of numerical worked out examples and problems some of which involve computer programs The Appendix gives some important data useful for solving problems in kinetics The book is designed for one semester course for undergraduate students of metallurgical discipline George Facer's Edexcel A Level Chemistry Student Book 1 George Facer, 2015-05-15 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Help higher achieving students to maximise their potential with a focus on independent learning assessment advice and model assessment answers in this new edition of George Facer's best selling textbook Encourages independent learning with notes and clear explanations throughout the content Strengthens understanding with worked examples of chemical equations and calculations Stretches the students with a bank of questions at the end of each chapter Provides assessment guidance and sample answers

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Chemical Kinetics: Beyond The Textbook Katja Lindenberg, Ralf Metzler, Gleb Oshanin, 2019-08-27 Processes

involving randomly moving particles which react either upon encounter or via distance dependent reaction rates are ubiquitous in nature A few stray examples are recombination of ions or holes and electrons excitation energy migration and quenching trapping of particles by other species coagulation binding of ligands and proteins to specific sites chemotaxis catalytically induced self propulsion polymerization growth of dendrites or aggregates or nuclei of a new phase Several decades ago it was recognized that the kinetic behavior in some systems with reactions and random transport is strongly affected by many factors which were not taken into account in previous studies These are to name but a few fluctuations in the spatial distributions of the reactants and fluctuations of the reactivity some essentially many particle phenomena effects of anomalous diffusion molecular crowding as well as the internal geometry of the reaction bath Within recent years along with a growing interest in chemical processes occurring in biological systems or cellular environments numerous advances have been made and considerable knowledge has been acquired These seminal contributions are however scattered among many journals and no attempt has been made so far to present a unified picture This book presents a general overview of different contemporary facets of chemical kinetics in a variety of different environments It includes 23 seminal works and reviews on different aspects of reaction processes in chemical physical and biophysical systems both theoretical and experimental

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