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Extra Practice with Inscribed and Central Angles section 10.4

Directions: Find each angle measure, arc measure, or value of the variable.

1. $m\angle XNM = \underline{40^\circ}$

$m\widehat{YX} = \underline{75^\circ}$



2. $m\widehat{TR} = \underline{150^\circ}$

$m\widehat{SR} = \underline{96^\circ}$



3. $m\widehat{WXY} = \underline{180^\circ}$

$m\angle WYX = \underline{50^\circ}$



4. $m\angle J = \underline{46^\circ}$

$m\angle K = \underline{90^\circ}$

$m\angle L = \underline{44^\circ}$



5. $m\widehat{BYD} = \underline{160^\circ}$

$m\angle YBC = \underline{88^\circ}$



6. $m\angle ECD = \underline{53^\circ}$

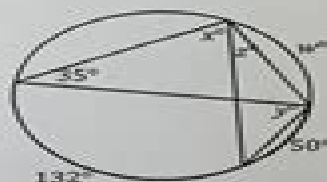


7. $x = \underline{66^\circ}$

$y = \underline{66^\circ}$

$z = \underline{26^\circ}$

$w = \underline{70^\circ}$



8. $m\angle F = \underline{115^\circ}$

$m\widehat{FH} = \underline{122^\circ}$



9. $x = \underline{3}$

$m\widehat{FGH} = \underline{238^\circ}$



$21x - 2 + 38x + 5 = 180$

$59x + 3 = 180$

$59x = 177$

$x = 3$

10. $m\angle W = \underline{22^\circ}$

$m\angle Y = \underline{52^\circ}$

$m\angle ZVY = \underline{74^\circ}$



$3x - 23 = 2x + 2$

$x = 25$

$(3x - 23)^\circ$

$(2x + 2)^\circ$

22°

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JS Bruner



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internuclear distances close to r_0 very close together at a distance that is less than the sum of the ionic radii several factors
contribute to the stability of ionic compounds

