

# AP® CHEMISTRY 2005 EXAMINATIONS CHEMISTRY SECTION

Question 10  
(8 points)

As sample of  $\text{C}_2\text{H}_2\text{Br}_4$  is subjected to continuous infrared light of 2.5 micrometers, continuous heat is produced. The product of this sample is subsequently subjected to gas-liquid chromatography.



The following PPM is recorded for the compound identified at temperature within 2.5 micrometers. (Heat decomposition used for known effect is not in this chromatogram.)

Chromatogram data:

- 10% in the temperature-increasing. Chromatogram data for compound is recorded. The retention time is approximately 10 minutes.
- 20% in the temperature-increasing. Chromatogram data for compound is recorded. The retention time is approximately 10 minutes.

1 point is earned for each correct answer.

$\text{C}_2\text{H}_2\text{Br}_4$  is a liquid with boiling point of 101.2°C. It is a colorless, odorless, non-flammable liquid.



The  $\text{C}_2\text{H}_2\text{Br}_4$  is a liquid with boiling point of 101.2°C. It is a colorless, odorless, non-flammable liquid. It is a liquid with boiling point of 101.2°C. It is a colorless, odorless, non-flammable liquid. It is a liquid with boiling point of 101.2°C. It is a colorless, odorless, non-flammable liquid.

The chromatogram data for the compound is recorded. The retention time is approximately 10 minutes. The chromatogram data for the compound is recorded. The retention time is approximately 10 minutes. The chromatogram data for the compound is recorded. The retention time is approximately 10 minutes.

1 point is earned for each correct answer.

# Response Question Ap Chemistry 2013

**Ying Liu**



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