Fluid and Electrolytes: Potter and Perry Chapter 42

- Overview
 - Fluid, electrolyte, and acid-base balances within the body maintain health and function of all body systems
 - <u>Fluid</u>: water that contains dissolved or suspended substances such as glucose, mineral salts, and proteins
 - Characteristics of body fluids influence body system function and have regulatory mechanisms that keep them in balance:
 - Fluid amount à volume
 - Concentration à osmolality
 - Composition à electrolyte concentration
 - Degree of acidity à pH
 - Water = 60% of adult body weight
 - Decreases with age
 - Fluid distribution: movement of fluid among its various compartments
 - Fluid located in 2 compartments:
 - Extracellular (ECF): outside of the cell
 - Intravascular liquid part of the blood (plasma)
 - Interstitial located between the cells and outside the blood vessels (within the tissues)
 - Intracellular (ICF): inside of the cell
 - Body tries to maintain homeostasis
 - Osmolality: measure of the # of particles per kilogram of water
 - Normal range: 285 295 mOsm/kg H2O
 - Tonicity: concentration of solutes in solution
 - Isotonic same tonicity as normal blood (no fluid shift)
 - Hypotonic more dilute than the blood
 - Hypertonic more concentration than normal blood
 - Electrolytes: a compound that separates into ions when it dissolves in water
 - Cations: + charged
 - Na, K, Ca, Mg
 - Anions: negatively charged
 - Cl. HCO3
- Movement of Water and Electrolytes:
 - Active transport; requires energy (ATP) to move electrolytes across cell
 membranes against concentration gradient (areas of low concentration to areas
 of high concentration)
 - Example: Sodium potassium pump
 - Moves Na out of a cell and K into it, keeping ICF lower in NA and higher in K than the ECF

Potter And Perry Chapter 42

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Potter And Perry Chapter 42:

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