

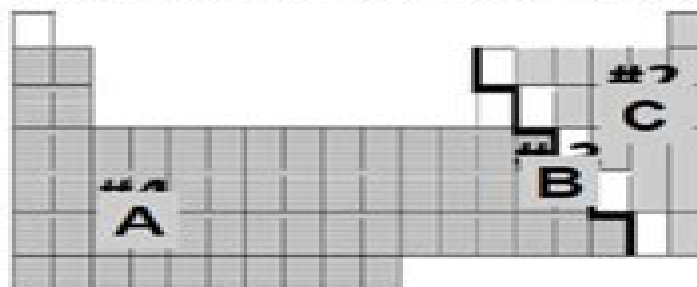
Name \_\_\_\_\_

Period \_\_\_\_\_

### Review: Unit 4 Periodic Table

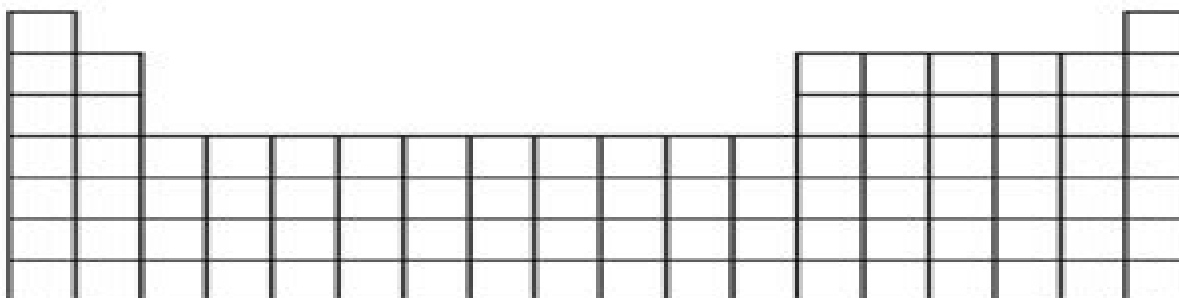
#### Part 1: History of the Periodic Table/ Families of the PT

1. Dimitri Mendeleev put elements in order based on \_\_\_\_\_
2. Henry Mosely put elements in order based on \_\_\_\_\_
3. Label the 3 sections of this periodic table as (Metalloid, Metals, Non-Metals)



#### Part 2: Periodic Trends

4. Label the periodic table with the following periodic table trends: Atomic Radius, Ionization Energy, Electronegativity (Use different colors for each trend)



5. Where on the Periodic Table is Electronegativity zero?

#### Part 3: Average Atomic Mass

6. What is the unit for average atomic mass? What does this unit mean, how is it measured?
7. 24.1% of all the isotopes of an element have a mass of 75.23 amu, 48.7% have a mass of 74.61 amu, and 27.2% have a mass of 75.20 amu. What is the average mass of this element?

# Review Of The Periodic Table Test Answer

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**Chemistry: Concepts and Problems** Clifford C. Houk, Richard Post, 1996-03-09

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