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CHAPTER 11 REVIEW

Molecular Composition of Gases

MIXED REVIEW

SHORT ANSWER Answer the following questions in the space provided.

- 1. _____ The average speed of a gas molecule is most directly related to the ____
 - (a) polarity of the molecule
 - (b) pressure of the gas
 - (c) temperature of the gas
 - (d) number of moles in the sample
- Which of the following statements best represents Graham's law for two gases at the same temperature?
 - (a) The molar masses and the velocities are inversely proportional.
 - (b) The molar masses and the velocities are directly proportional.
 - (c) The molar masses and the square roots of the velocities are directly proportional.
 - (d) The molar masses and the squares of the velocities are inversely proportional.
- 3. Water is synthesized from its elements according to the following equation:

$$2H_2(g) + O_2(g) \rightarrow 2H_2O(l)$$

- a. Which is greater, the volume of H₂ consumed or the volume of O₂ consumed?
- b. Which is greater, the mass of H₂ consumed or the mass of O₂ consumed?
- 4. Consider a 2.24 L sample of an ideal gas at STP.
 - _____ a. How many moles are present in the sample, or is more data needed?
 - b. How many molecules are present in the sample, or is more data needed?
 - e. What is the mass of the sample, or is more data needed?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

Calculate the volume of 3.5 mol of an ideal gas at 24°C and 760 torr.

Review Molecular Composition Of Gases

Rosina Ehmann

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William Crookes, James H. Gardiner, Gerald Druce, H. W. Blood-Ryan, 1877 Chemical Abstracts, 1922 Chemical News and Journal of Industrial Science of Industrial Science, 1905 Engineering and Boiler House Review, 1908 Quantum Mechanics

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Fuel Economy Review ,1929 The Old and New Persistent Organic Pollutants Ravi Naidu,2025-08-01 Our understanding of persistent organic pollutants POPs their exposure pathways and their impact on the environment and human health is constantly evolving and the list of new and emerging POPs is constantly changing This book provides a comprehensive coverage of new and old hazardous chemicals their physical and chemical properties their breakdown products their fate in the environment and the environmental and human risk impact It discusses global policies based on the United Nations FAO frameworks explains the severity of contamination and raises awareness on the assessment and remediation of contaminated sites in developed and developing countries Features Provides a broad temporal perspective on POPs with contributions from a global team of experts Covers chemistry toxicology remediation regulation and conventions related to POPs Explains systematically the fate and behavior of POPs and their effect on the environment and ultimately the impact on human health Brings together for the first time information on global policies on POPs Includes case studies that detail assessment criteria of old and new POPs as well as remediation technologies This book is an excellent resource for

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Energy Research and Development Administration,1977 Chemical News and Journal of Physical Science ,1887

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