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THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct answer next to the item.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed amount of gas at constant pressure
2. The volume of a fixed amount of gas at constant pressure as the absolute temperature is changed
3. The ratio of the mass of a gas to its volume, independent of the actual amount of gas used
4. Temperature at which a substance changes from solid to liquid
5. A measure of how much the volume of a fixed amount of gas changes
6. The a parameter of gas pressure dependence for compressing gas nonreversibly with pressure
7. Temperature at which a solid changes directly to liquid
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of how much a fixed amount of gas expands
10. Solid to liquid transition

Column II

- a. constant pressure
- b. ideal gas constant R
- c. volume
- d. condensation
- e. Boyle's law
- f. Boyle's constant
- g. constant non-reversible pressure
- h. volume
- i. compressibility
- j. condensation of vapors

B. Multiple Choice

Choose the correct answer and write its letter on the line.

1. Subliming the solvent of a solid-solvent system from a solid solvent leaving nonvolatile solute behind is
 - a. the sublimation technique
 - b. the method of zone refining
 - c. the distillation technique
 - d. the method of zone refining

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