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## Chapter 17 & 19 Study Guide

### **Key Terms**

*Directions: Define each of the following key terms from chapters seventeen and nineteen on a separate sheet of paper.*

- **Atom** - the smallest piece of matter that still has the properties of the element
- **atomic number** – number of protons in an atom's nucleus
- **atomic mass** – average mass of all an element's isotopes
- **ductile** – capable to being drawn into wire
- **electron** – subatomic particle with a negative charge
- **electron cloud** – area outside the nucleus in which electrons can be found.
- **Electron dot diagram** – uses the elements symbol and dots to represent the valance electrons in an element.
- **energy level** – areas within the electron cloud where electrons are most likely to be found.
- **Family** – vertical columns of the periodic table
- **Group** - vertical columns of the periodic table
- **half-life** – the amount of time it take half the sample of a radioactive isotope to decay.
- **Isotope** – atoms of the same element that have different number of neutrons.
- **Lustrous** – having a shiny appearance
- **Malleable** – can be hammered flat or rolled into sheets
- **mass number** – sum of protons and neutrons in the nucleus of an atom.
- **Metal** – element that typically is hard, shiny, solid, malleable, and is good conductor of heat and electricity.
- **Neutron** – neutrally charged subatomic particle.
- **Nonmetal** - element that typically is brittle solid or gas, not shiny, not malleable, and is poor conductor of heat and electricity.
- **Nucleus** – center of the atom
- **Period** – horizontal row of the periodic table.
- **periodic table** – organized list of all know elements.
- **Proton** – positively charged subatomic particle
- **transition metal** – Elements found in families 3-12 of the periodic table.
- **transuranium element** – an elements containing more than 92 protons in its nucleus
- **quark** – particle believed to make up protons and neutrons.
- **valence electron** – electrons found in the outermost energy level of an atom.

# Physics Chapter 17 Study Guide Answers

**Victor M. Corman**



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## **Table of Contents Physics Chapter 17 Study Guide Answers**

1. Understanding the eBook Physics Chapter 17 Study Guide Answers
  - The Rise of Digital Reading Physics Chapter 17 Study Guide Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Physics Chapter 17 Study Guide Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Physics Chapter 17 Study Guide Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Chapter 17 Study Guide Answers

- Personalized Recommendations
- Physics Chapter 17 Study Guide Answers User Reviews and Ratings
- Physics Chapter 17 Study Guide Answers and Bestseller Lists
- 5. Accessing Physics Chapter 17 Study Guide Answers Free and Paid eBooks
  - Physics Chapter 17 Study Guide Answers Public Domain eBooks
  - Physics Chapter 17 Study Guide Answers eBook Subscription Services
  - Physics Chapter 17 Study Guide Answers Budget-Friendly Options
- 6. Navigating Physics Chapter 17 Study Guide Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Physics Chapter 17 Study Guide Answers Compatibility with Devices
  - Physics Chapter 17 Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Physics Chapter 17 Study Guide Answers
  - Highlighting and Note-Taking Physics Chapter 17 Study Guide Answers
  - Interactive Elements Physics Chapter 17 Study Guide Answers
- 8. Staying Engaged with Physics Chapter 17 Study Guide Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Physics Chapter 17 Study Guide Answers
- 9. Balancing eBooks and Physical Books Physics Chapter 17 Study Guide Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Physics Chapter 17 Study Guide Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Physics Chapter 17 Study Guide Answers
  - Setting Reading Goals Physics Chapter 17 Study Guide Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physics Chapter 17 Study Guide Answers

- Fact-Checking eBook Content of Physics Chapter 17 Study Guide Answers
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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