

Electrons in Atoms

Section 5.1 Light and Quantized Energy

In your textbook, read about the wave nature of light.

Use each of the terms below just once to complete the passage.

amplitude
light

energy
wave

frequency
wavelength

hertz
speed

Electromagnetic radiation is a kind of **(1)** _____ that behaves like a(n) **(2)** _____ as it travels through space. **(3)** _____ is one type of electromagnetic radiation. Other examples include X rays, radio waves, and microwaves.

All waves can be characterized by their wavelength, amplitude, frequency, and **(4)** _____. The shortest distance between equivalent points on a continuous wave is called a(n) **(5)** _____. The height of a wave from the origin to a crest or from the origin to a trough is the **(6)** _____. **(7)** _____ is the number of waves that pass a given point in one second. The SI unit for frequency is the **(8)** _____, which is equivalent to one wave per second.

Use the figure to answer the following questions.



9. Which letter(s) represent one wavelength? _____
10. Which letter(s) represent the amplitude? _____
11. If twice the length of A passes a stationary point every second, what is the frequency of the wave?

Prentice Hall Chemistry Exam Answers Chapter 5

JA Banks



Prentice Hall Chemistry Exam Answers Chapter 5:

This book delves into Prentice Hall Chemistry Exam Answers Chapter 5. Prentice Hall Chemistry Exam Answers Chapter 5 is an essential topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Prentice Hall Chemistry Exam Answers Chapter 5, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Prentice Hall Chemistry Exam Answers Chapter 5
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