

14

THE BEHAVIOR OF GASES

Chapter Test II

A. Matching

Match a word or phrase in Column I with the correct description in Column II. Write the letter of the correct word or phrase in the space.

Column I

1. The constant relating unit temperature change to the change in volume of a fixed mass of gas at constant pressure
2. The volume of a fixed mass of gas at constant pressure as the absolute temperature is changed
3. The ratio of the mass of a gas to its volume, independent of the state of the substance
4. Temperature at which the average kinetic energy of the particles of a gas is zero
5. A measure of how much the volume of a fixed mass of gas changes
6. The a combination of gas pressure, temperature, and volume of the gas mathematically expressed
7. An ideal gas that obeys the gas laws perfectly
8. $\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$
9. A measure of how much a gas differs from an ideal gas
10. A gas at zero pressure

Column II

- a. pressure gradient
- b. ideal gas constant R
- c. density
- d. extrapolation
- e. Boyle's law
- f. Boyle's equation
- g. Charles law/Charles's equation
- h. diffusion
- i. compressibility
- j. Dalton's law of partial pressures

B. Multiple Choice

Choose the correct answer and write its letter in the space.

1. According to Graham's law of effusion, the rate of effusion of a gas is inversely proportional to the _____ of the gas.
a. the square root of its density
b. the square of its mass
c. the square root of its mass
d. the square of its density

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CL Gary



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