

2-4: Precipitation Reactions

1. Start *Virtual ChemLab*, select *Reactions and Stoichiometry*, and then select *Precipitation Reactions* from the list of assignments. The lab will open in the Inorganic laboratory.
2. React each of the cations (across the top) with each of the anions (down the left) according to the data table using the following procedures:

Data Table

	AgNO_3 (Ag^+)	$\text{Pb}(\text{NO}_3)_2$ (Pb^{2+})	$\text{Ca}(\text{NO}_3)_2$ (Ca^{2+})
Na_2CO_3 (CO_3^{2-})	<i>a</i>	<i>f</i>	<i>k</i>
Na_2S (S^{2-})	<i>b</i>	<i>g</i>	<i>l</i>
NaOH (OH^-)	<i>c</i>	<i>h</i>	<i>m</i>
Na_2SO_4 (SO_4^{2-})	<i>d</i>	<i>i</i>	<i>n</i>
NaCl (Cl^-)	<i>e</i>	<i>j</i>	<i>o</i>

- a. Drag a test tube from the box and place it on the metal test tube stand. You can then click on the bottle of Ag^+ ion solution on the shelf to add it to the test tube.
 - b. Click on the **Divide** button on the edge of the lab bench four times to make four additional test tubes containing Ag^+ . With one test tube in the metal stand and four others in the blue rack, click on the Na_2CO_3 bottle on the reagent shelf and observe what happens in the window at the top left. Record your observation in the table above. If the solution remains clear, record NR for no reaction. Drag this test tube to the red disposal bucket on the right.
 - c. Drag a second tube from the blue rack to the metal stand. Add Na_2S , record your observations and discard the tube. Continue with the third, fourth and fifth tube, but add NaOH , Na_2SO_4 , and NaCl respectively. Record your observations and discard the tubes. When you are finished, click on the red disposal bucket to clear the lab.
 - d. Repeat steps a-c for five test tubes of Pb^{2+} and Ca^{2+} . Record your observations in the data table. If no precipitate forms write NR for no reaction.
3. What happens in grid space *d*? What other reactions give similar results? Is it necessary to write an equation when no reaction occurs? Explain. _____

 4. Write balanced equations for all precipitation reactions you observed.
 - a.
 - b.
 - c.

Precipitation Reactions Virtual Chemlab Answers

Jianjun Gao



Precipitation Reactions Virtual Chemlab Answers:

Decoding **Precipitation Reactions Virtual Chemlab Answers**: Revealing the Captivating Potential of Verbal Expression

In a time characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Precipitation Reactions Virtual Chemlab Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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