

| Term              | Definition                                                                                                        |
|-------------------|-------------------------------------------------------------------------------------------------------------------|
| activation energy | the minimum amount of energy molecules must have to undergo reactions                                             |
| allotrope         | different structural forms of the same element with different chemical properties. Examples: diamond, graphite    |
| allotropy         | property of substances that exist in different structural forms in the same state                                 |
| allotrope         | property of substances that exist in different structural forms in the same state                                 |
| alloy             | metallic material made by dissolving a substance that has a smaller atomic number, or replacing some of the atoms |
| alloy             | one or more metals or metalloids in solution                                                                      |

# Plato Chemistry Semester

**West Virginia University**

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Superintendent of Public Instruction, 1892      **Biennial Report of the Superintendent of Public Instruction of the State of Washington** Washington (State). Superintendent of Public Instruction, 1892      **Computers in Chemical**

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